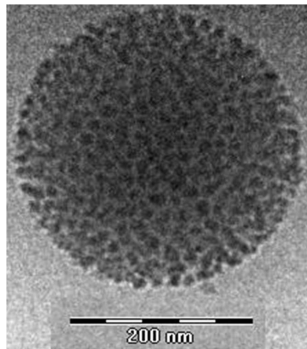


# Formulating Nano-particles into Robust Particle-Stabilized Emulsions

Karin Persson, Isabel Mira, Adam Feiler, Anna Millqvist-Fureby, Jonas Gustafsson

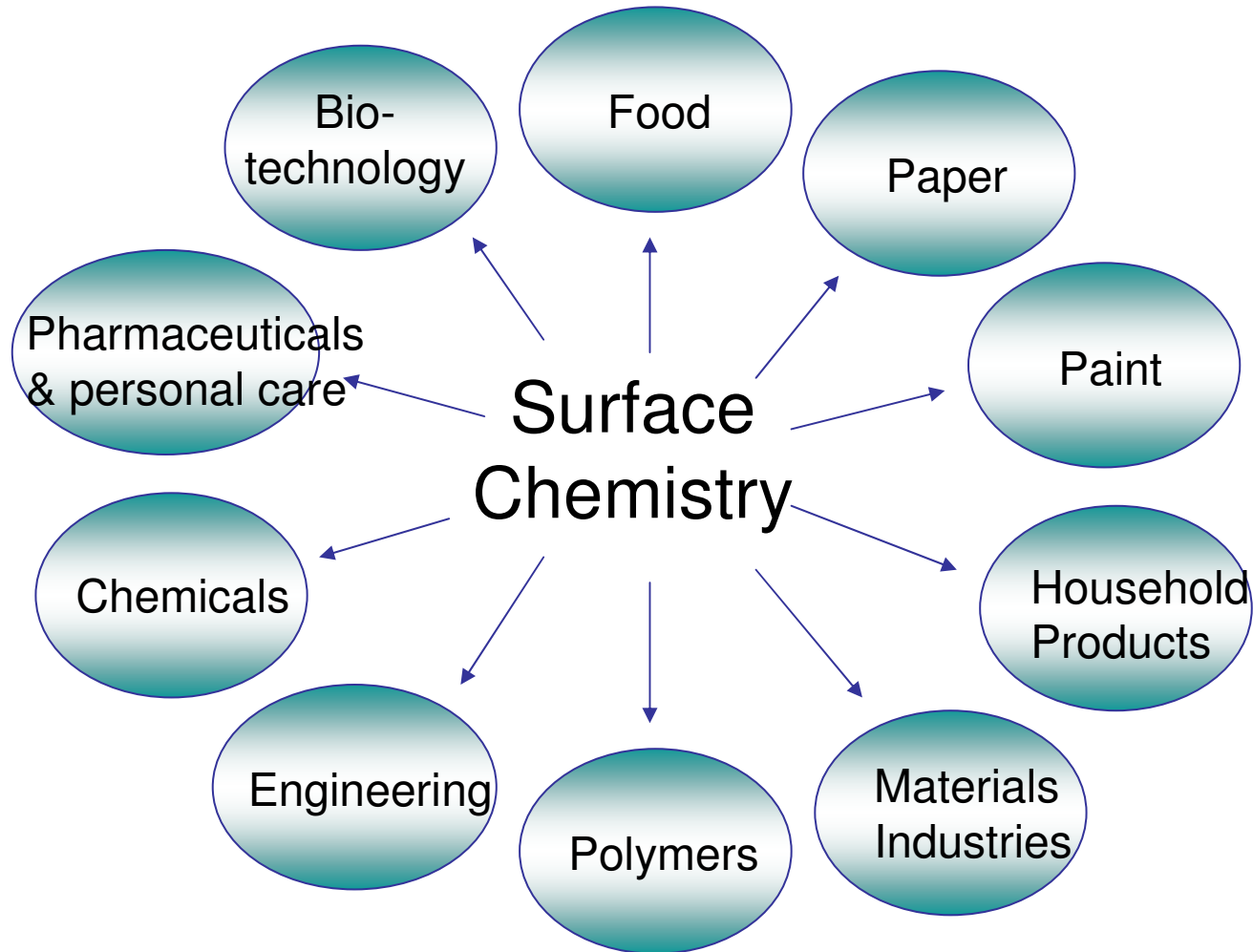
YKI, Institute for Surface Chemistry  
*karin.persson@yki.se*



- Particle Stabilized Emulsions
- Preparation of Particle Stabilized O/W Nanoemulsions
- Stability of PS Nanoemulsions.
- Spray-drying of PS Nanoemulsions

# YKI : your R&I Partner in Applied Surface Chemistry

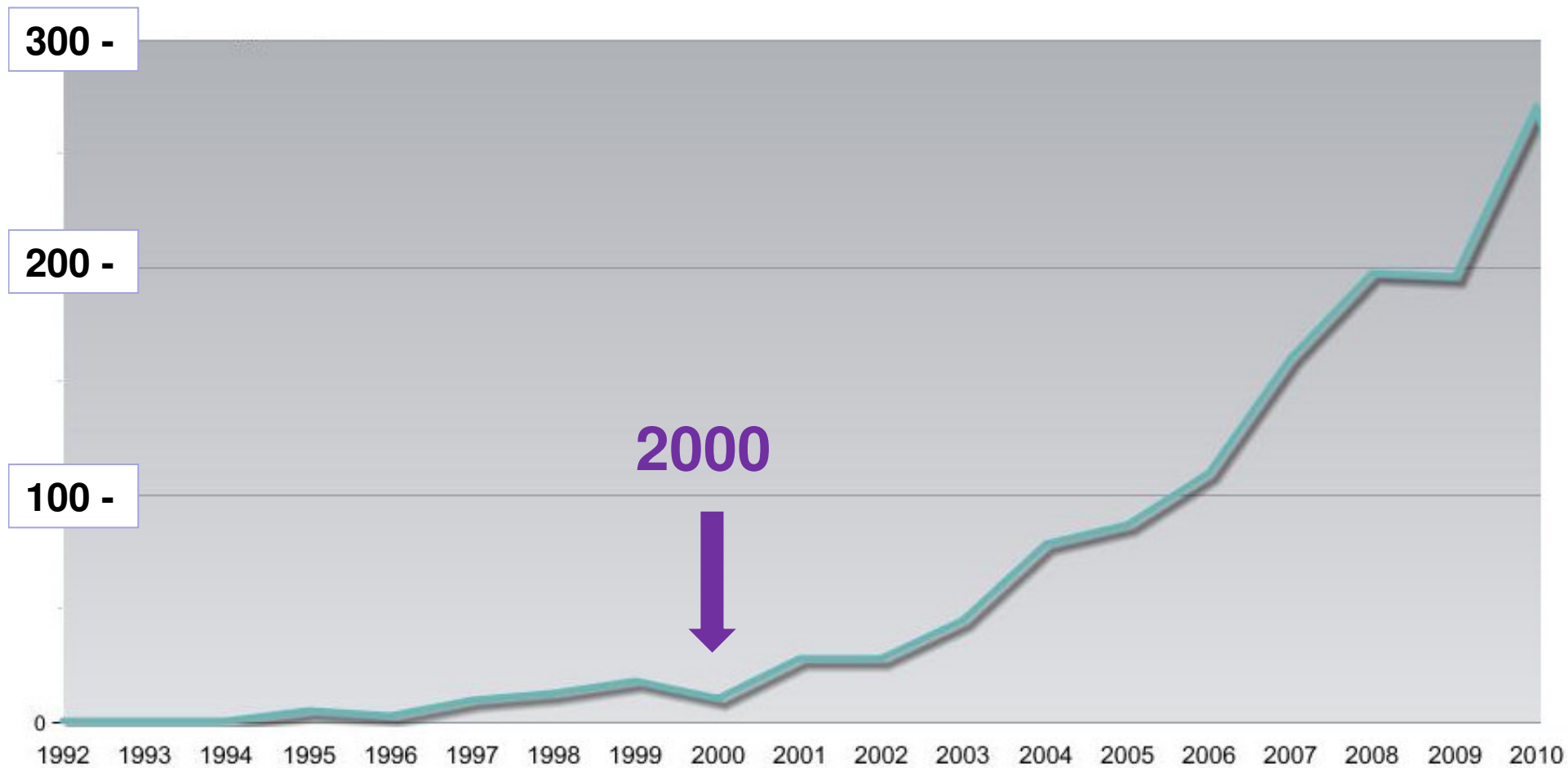
---



# Some of the Member Companies.....

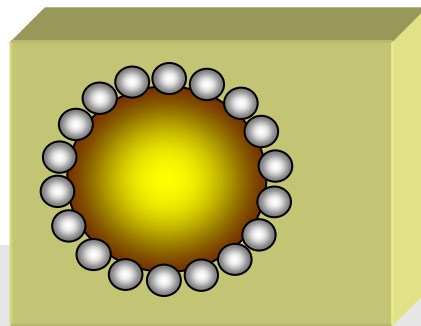
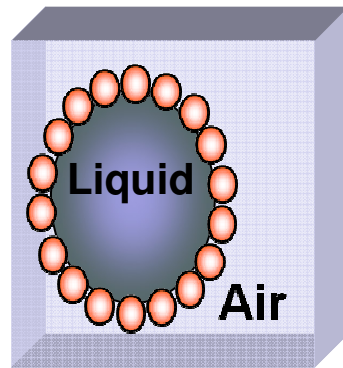
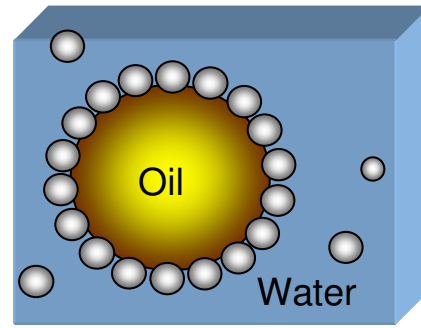
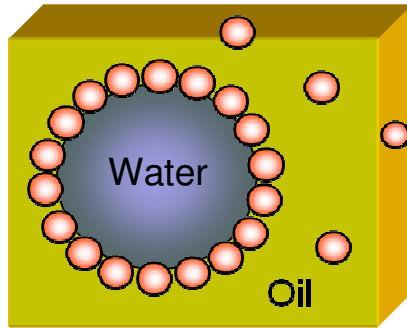


# Pickering and Particle Stabilized Article Publishing Trends



Source: Thomson Innovation®, [www.thomsoninnovation.com](http://www.thomsoninnovation.com)

# Particle Stabilized Structures



## *Types of emulsions*

*W/O and O/W/O*

*O/W and W/O/W*

*Liquid marbles*

*Dry water*

*Microcapsules*

*Colloidosomes*

## **Applications**

*Nano-particle design*

*Synthesis*

*Nanocomposites*

*Porous materials*

*Nano-particle vehicles*

*Alternative penetration*

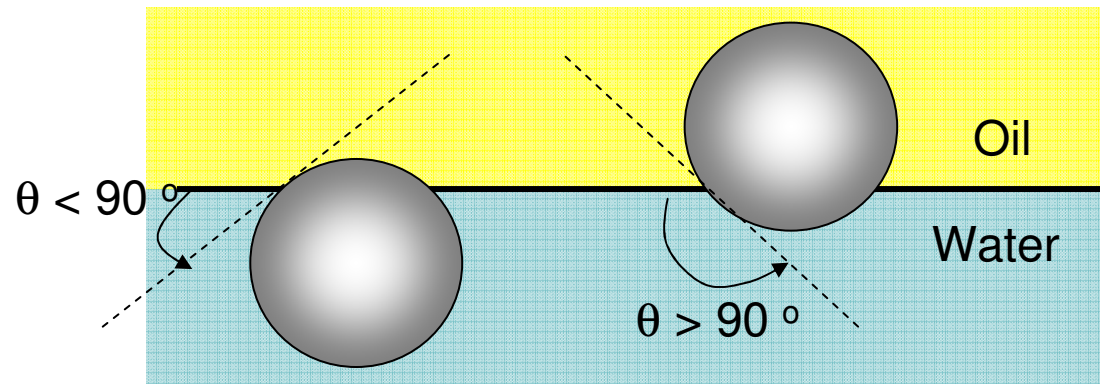
*profiles*

*New delivery systems*

# Particle Stabilized Emulsions: Formulation

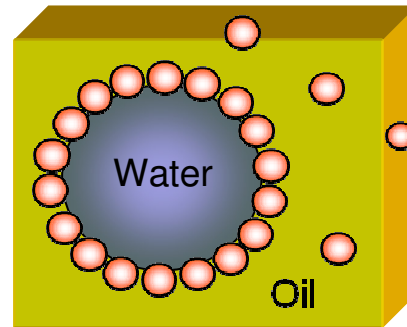
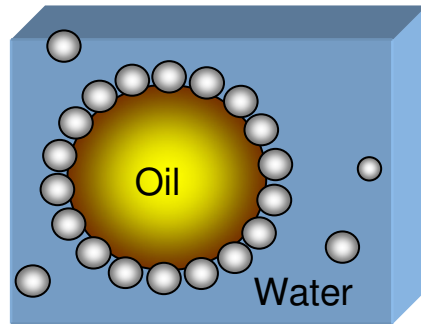
## I. Physicochemical Aspects

Particle wettability determines preferred emulsion type



Wettability determined by:

- ▶ Nature of the particle
- ▶ Nature of the oil and aqueous phase



Binks, B.P., *Curr. Opin. Colloid Interface Sci.* 7, 21-41 (2002).

# Particle Stabilized Emulsions: Formulation

**E proportional to  $R^2 \rightarrow$**

Larger particles are generally more effective

**For 90 ° contact angle:**

1  $\mu\text{m}$  particle  $E \sim 10^7 \text{ kT}$

10 nm particle  $E \sim 10^3 \text{ kT}$

1 nm particle  $E \sim 10 \text{ kT}$

Globular proteins  $E \sim \text{a few kT}$

Surfactants  $E \sim 1 \text{ kT}$

**Stabilization energy of a particle**

Energy required to remove particle from oil-water interface is given by

$$-E = \pi r^2 \gamma (1 \pm \cos \theta)^2$$



Pieranski, P. *Physical Review Letters* **1980**, 45, 569-572.

Binks, B. P.; Lumsdon, S. O., *Langmuir* **2001**, 17, 4540-4547.

# Other Variables Affecting the Outcome of Emulsification

---

## II. Composition

- Amount
- Ratios
- Order of addition

## III. Emulsification protocol

- Shear (turbulence)
- Mixer geometry (turbine, blade)
- Energy (intensity, duration)



**...the properties of an emulsion depend on  
a complex interplay of factors!**

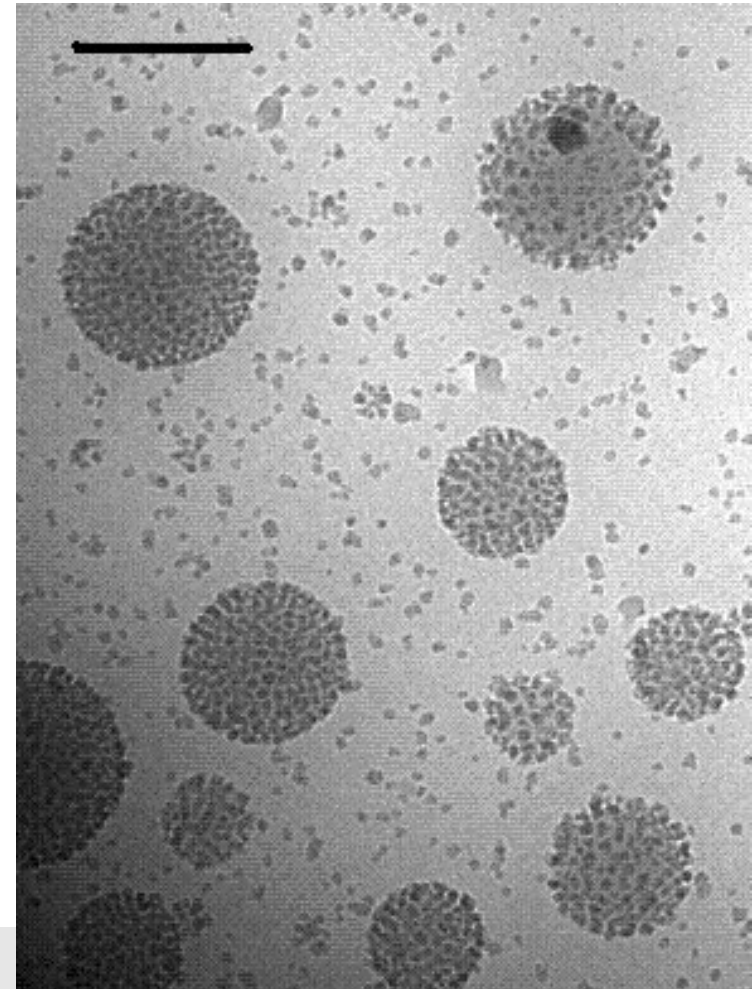


# Novel Development: Robust Particle-stabilized Nanoemulsions

---

- Colloidal Pickering emulsions
- Ultra-stable droplets
  - *Temperature, electrolyte, surfactants, etc*

200 nm

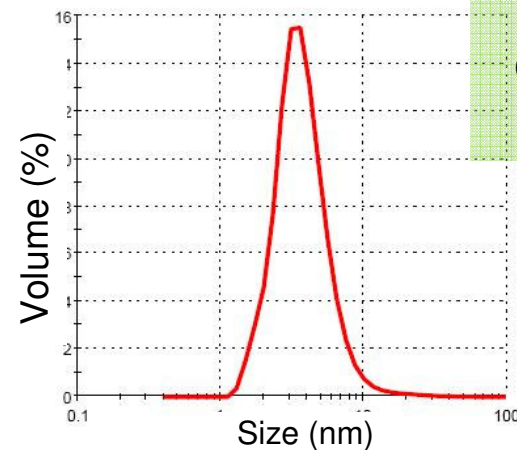
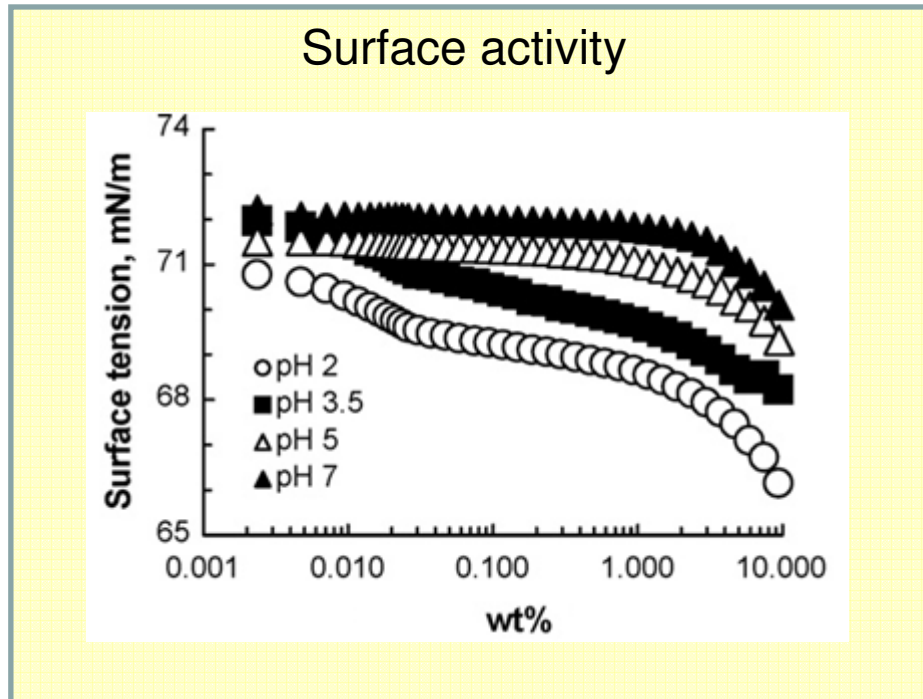


# The Particles: Commercial Silica Sol

Silica Sol



- ▶ Hydrophobically modified (methoxysilanated or ethoxysilanated) silica particles
- ▶ Aqueous dispersion ( pH 7.5, 30% w)
- ▶ Particle diameter: 7 nm



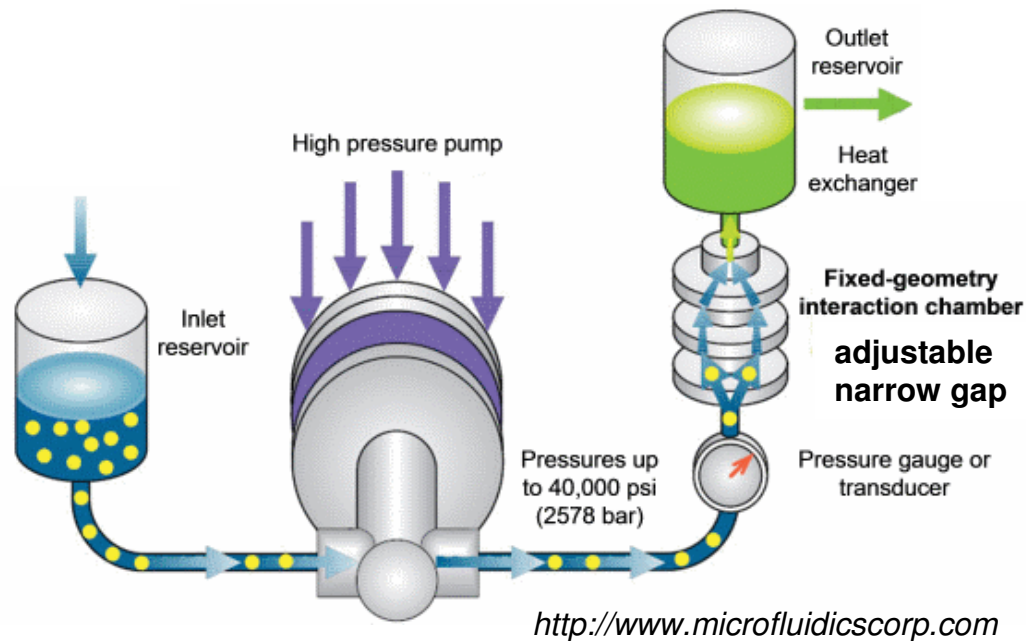
Stabilization of O/W emulsions

Size distribution by volume as determined by dynamic light scattering

# Preparation of Particle Stabilized O/W Nanoemulsions

## *Emulsification Protocol*

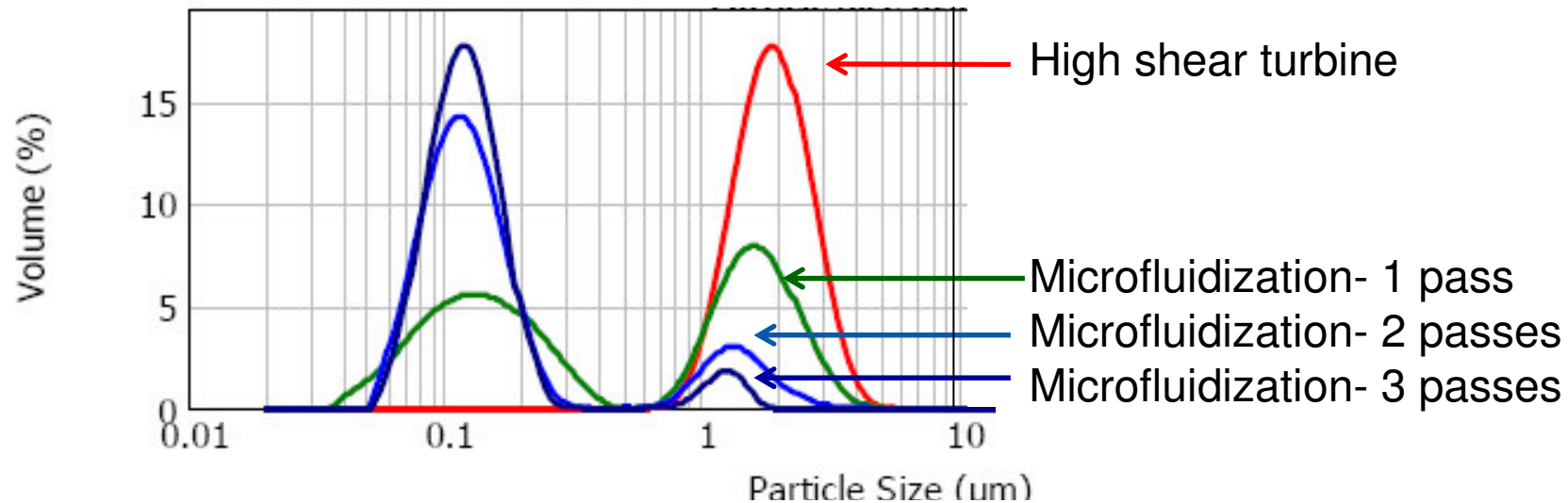
- ▶ Ultra shear device - High pressure homogenizer type
- ▶ Production of extremely fine emulsion (drops in nanometer size range)



# Preparation of Particle Stabilized O/W Nanoemulsions

## Emulsification Protocol Studies

Oil phase: Hexadecane (C16)  
Particle concentration: 30% w/v

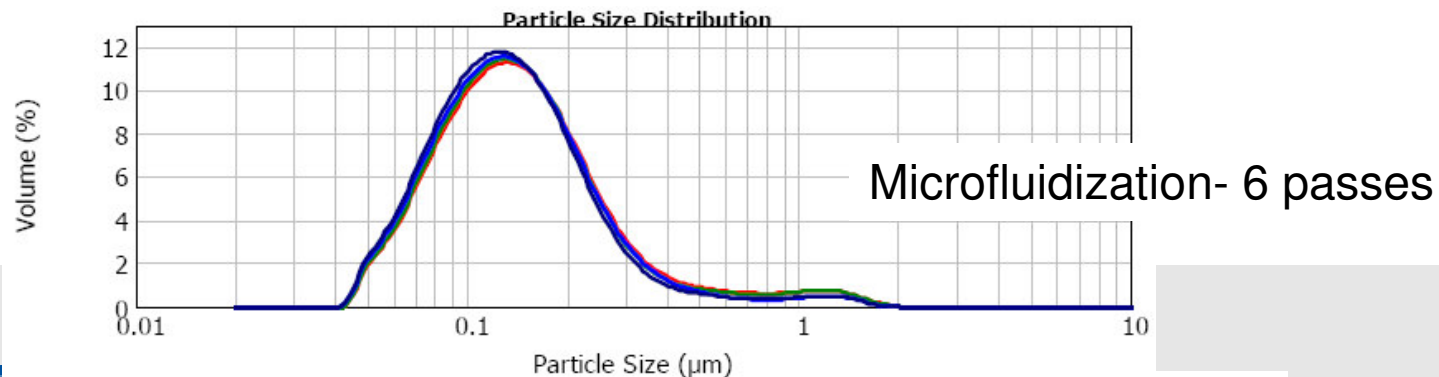
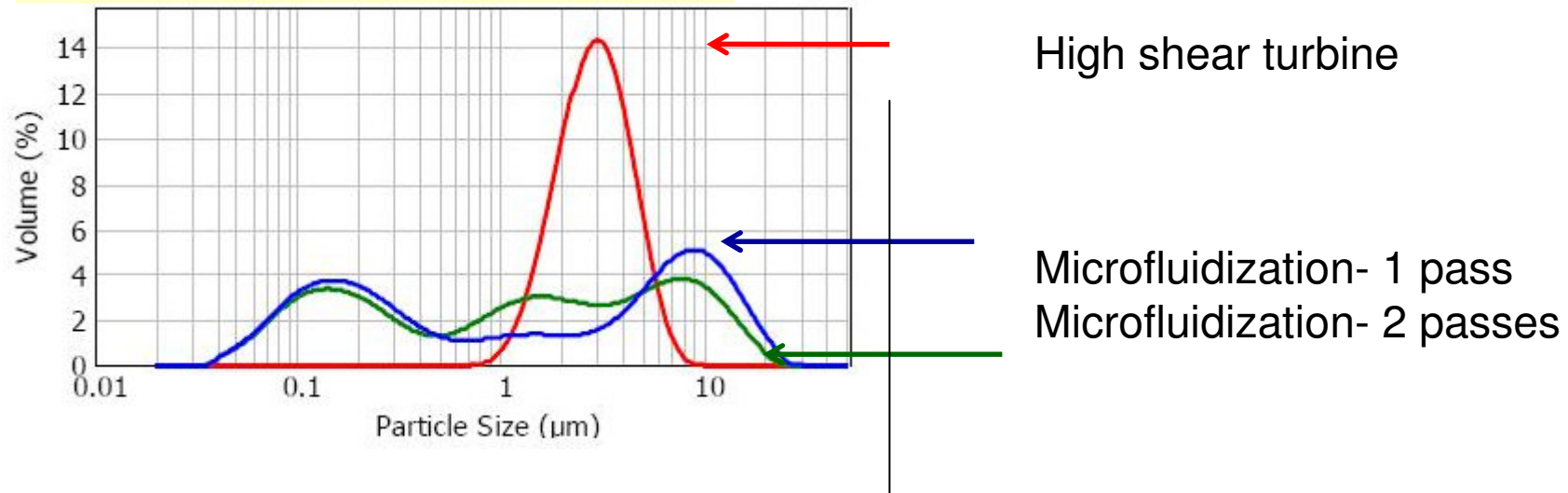


*Particle size distribution of freshly prepared emulsions*

# Preparation of Particle Stabilized O/W Nanoemulsions

## Emulsification Protocol Studies

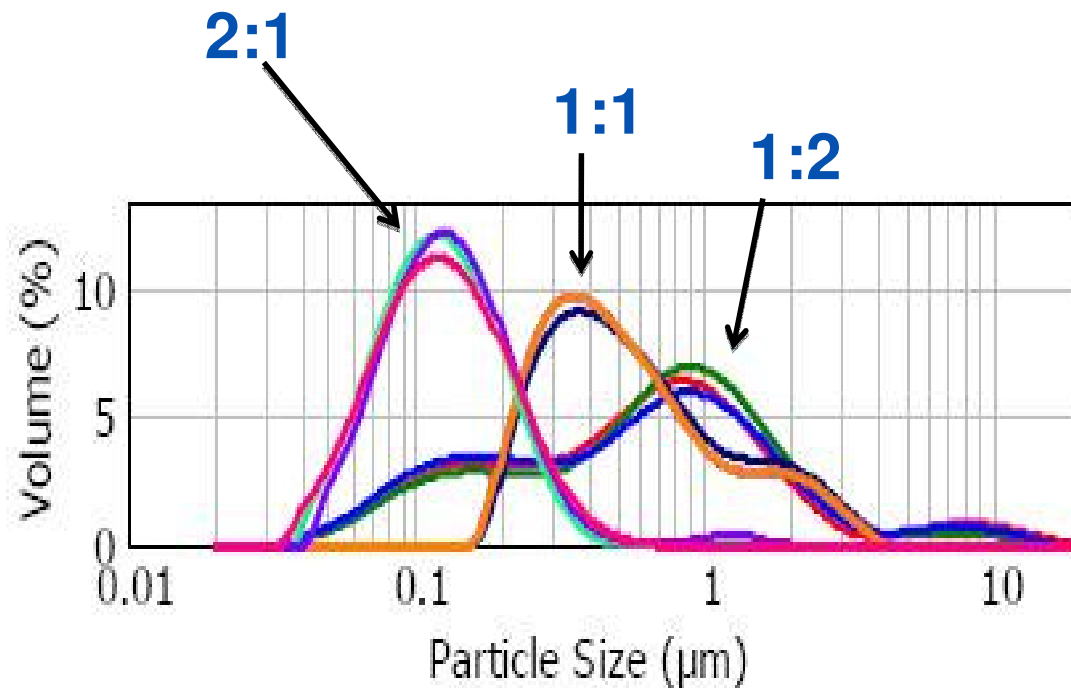
Oil phase: Hexadecane (C16)  
Particle concentration: 10% w/v



Particle size distribution of freshly prepared emulsions

# Nanoemulsions of Food Oils

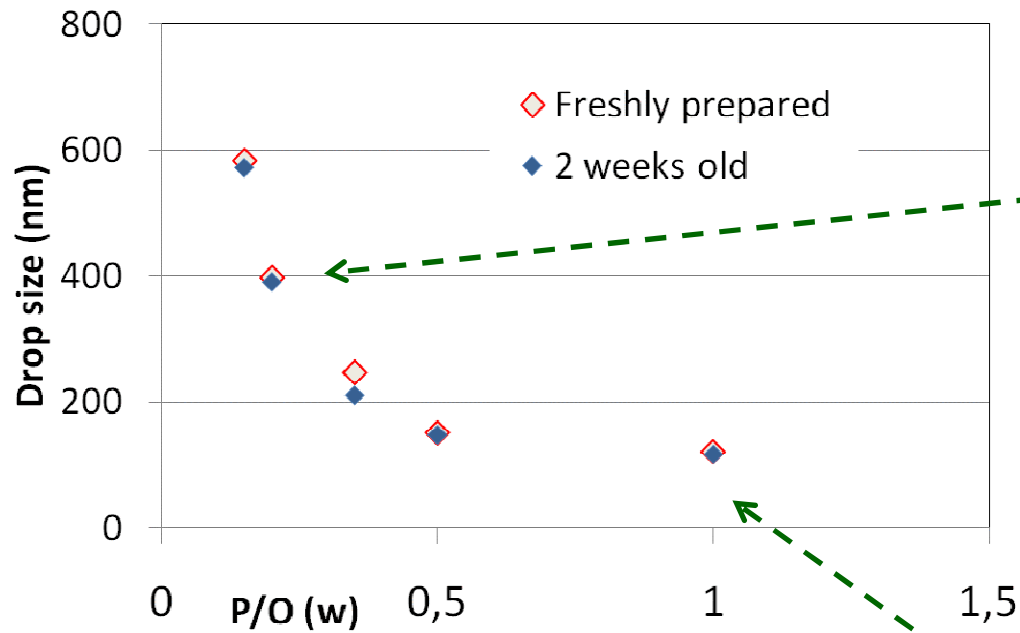
## Effect of particle : rapeseed oil ratio



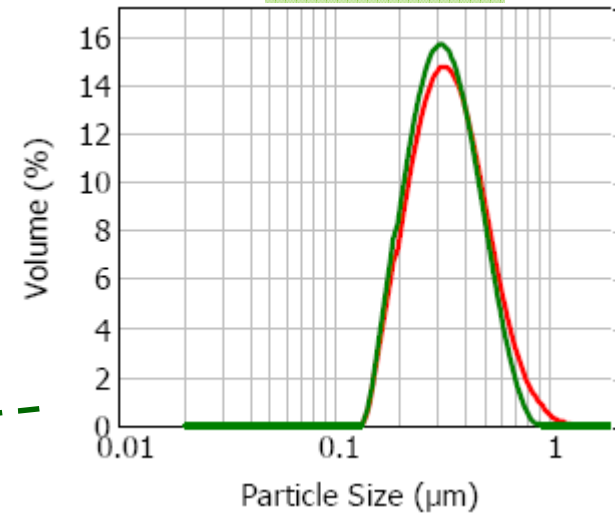
Droplet size as a function of processing time at 4.5 bar pressure.  
↑ 5 ,10 and 15 minutes for the systems

# Stability Studies: Time

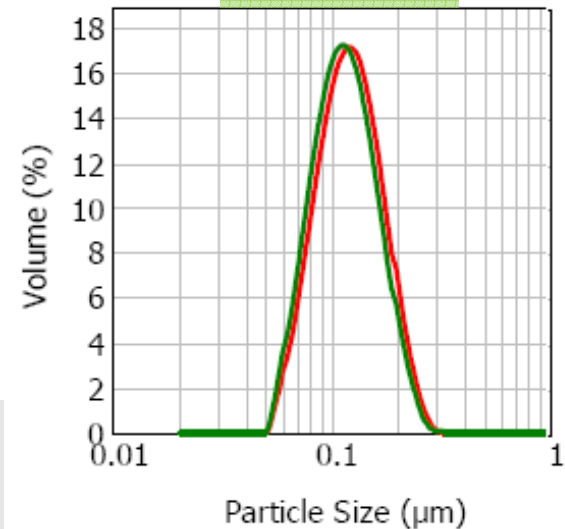
Oil phase: Squalene (10% w/v)  
P/O ratio: 0.15-1



3 months

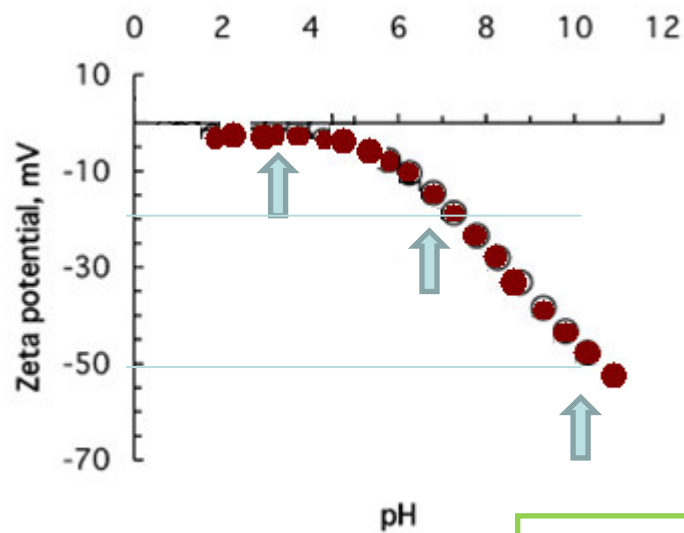


5 months

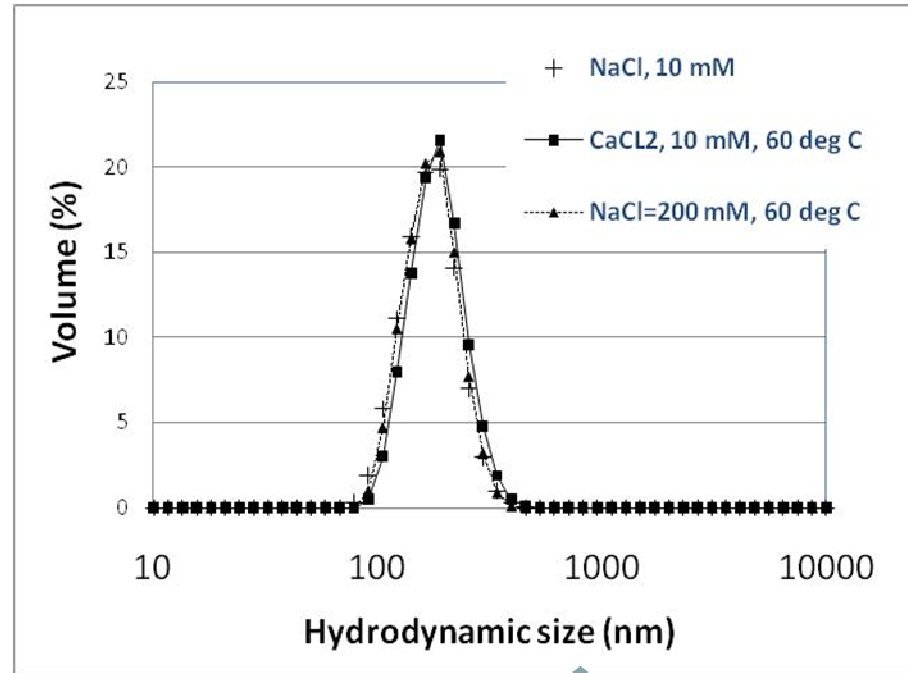


# Stability Studies: Temperature, pH, Surfactants and Electrolytes

	SDS	Tween 80	Dotab
pH	1 w/w%	1 w/w%	1 w/w%
3			
7			
10			



	CaCl <sub>2</sub>
pH	20 mM
3	
7	
10	



*Preserved narrow droplet distribution after several weeks of exposure to electrolytes & elevated temperatures*

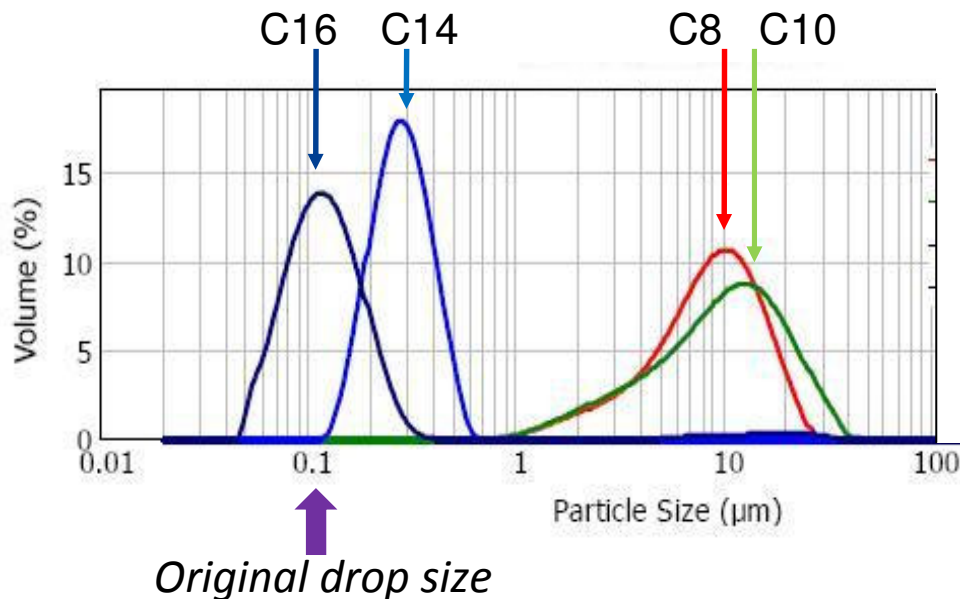


# Low Solubility of Oil in Continuous Phase Needed to Reduce Ostwald Ripening

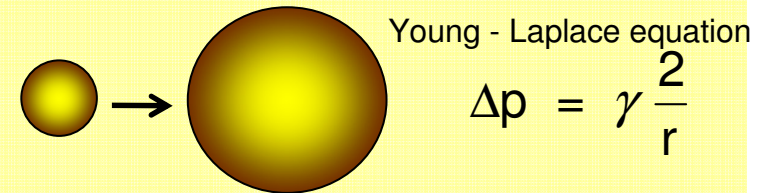
*Droplet size distribution after 2 weeks storage*

Oil phase: variable  
 Linear alkanes (C8, C10, C14, C16)  
 P/O ratio: 1.2 (10% w/v)

## ► Effect of Oil Phase

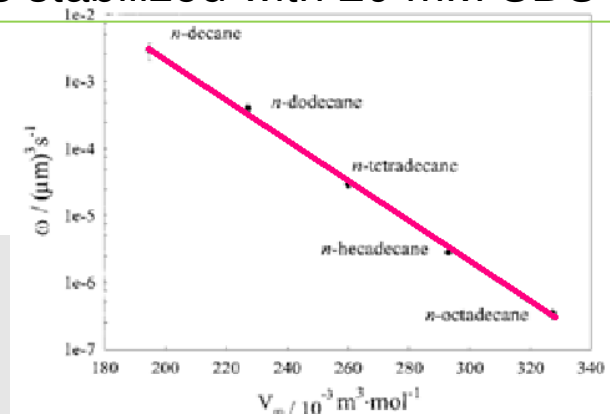


## Ostwald Ripening

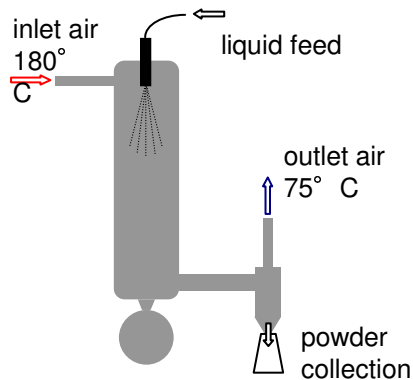


- Molecules of the dispersed phase travel from the small droplets to the larger ones.
- Leads to emulsion breakage

Ostwald ripening rate of linear alkane emulsions stabilized with 20 mM SDS



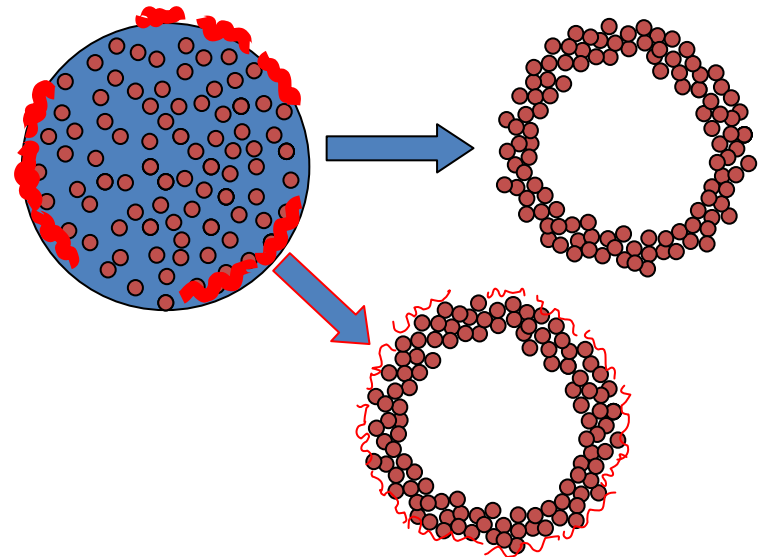
# Surface Formation in Spray-dried Suspensions



Mass transport by diffusion and convection

Moisture moves from the interior of the droplet by diffusion and capillary forces

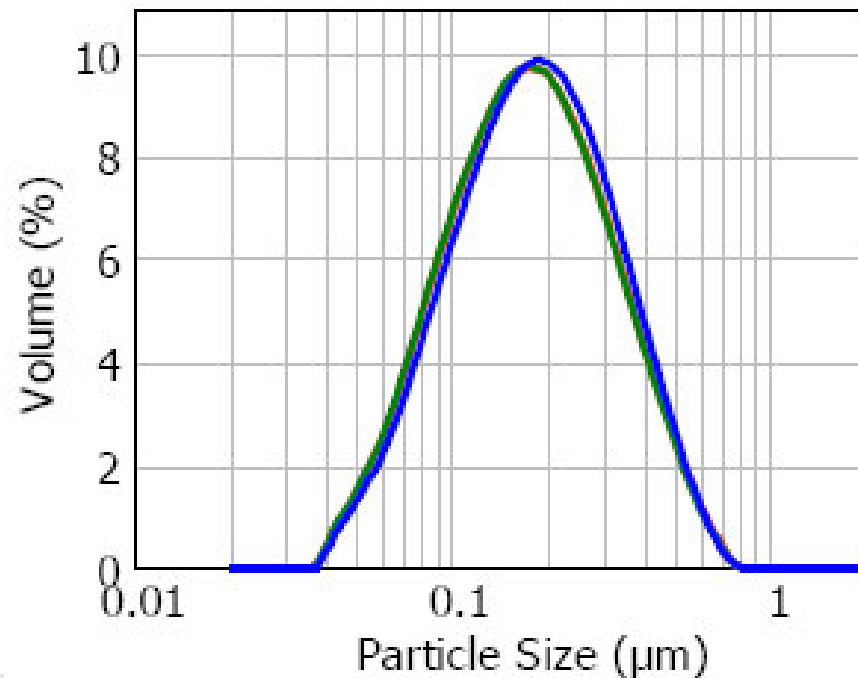
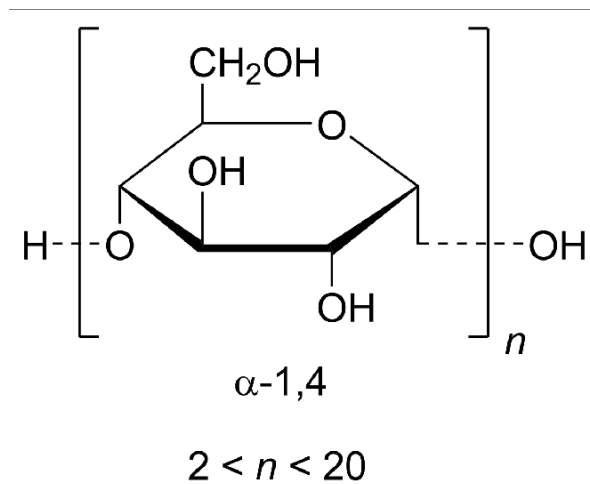
- Solid primary particles
- Binder
- Flocculant
- Non-adsorbed binder may migrate to surface and form a film



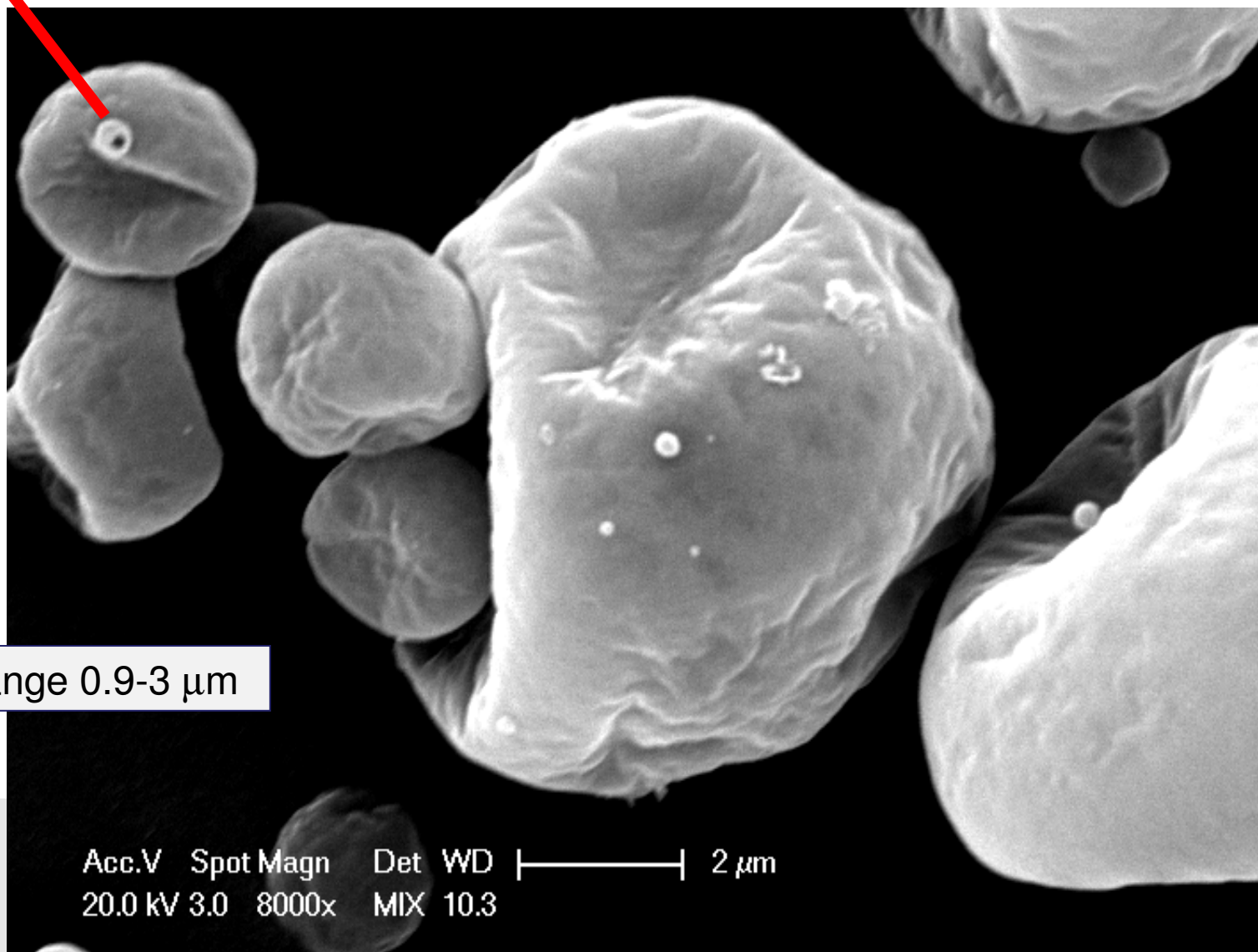
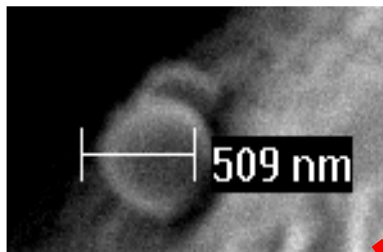
# Spray-drying of Pickering Emulsions

Addition of maltodextrin binder did not change droplet size of original emulsion

10% hexadecane,  
3.5% silica particles  
3.5% maltodextrins



# SEM of Spray-dried Emulsion

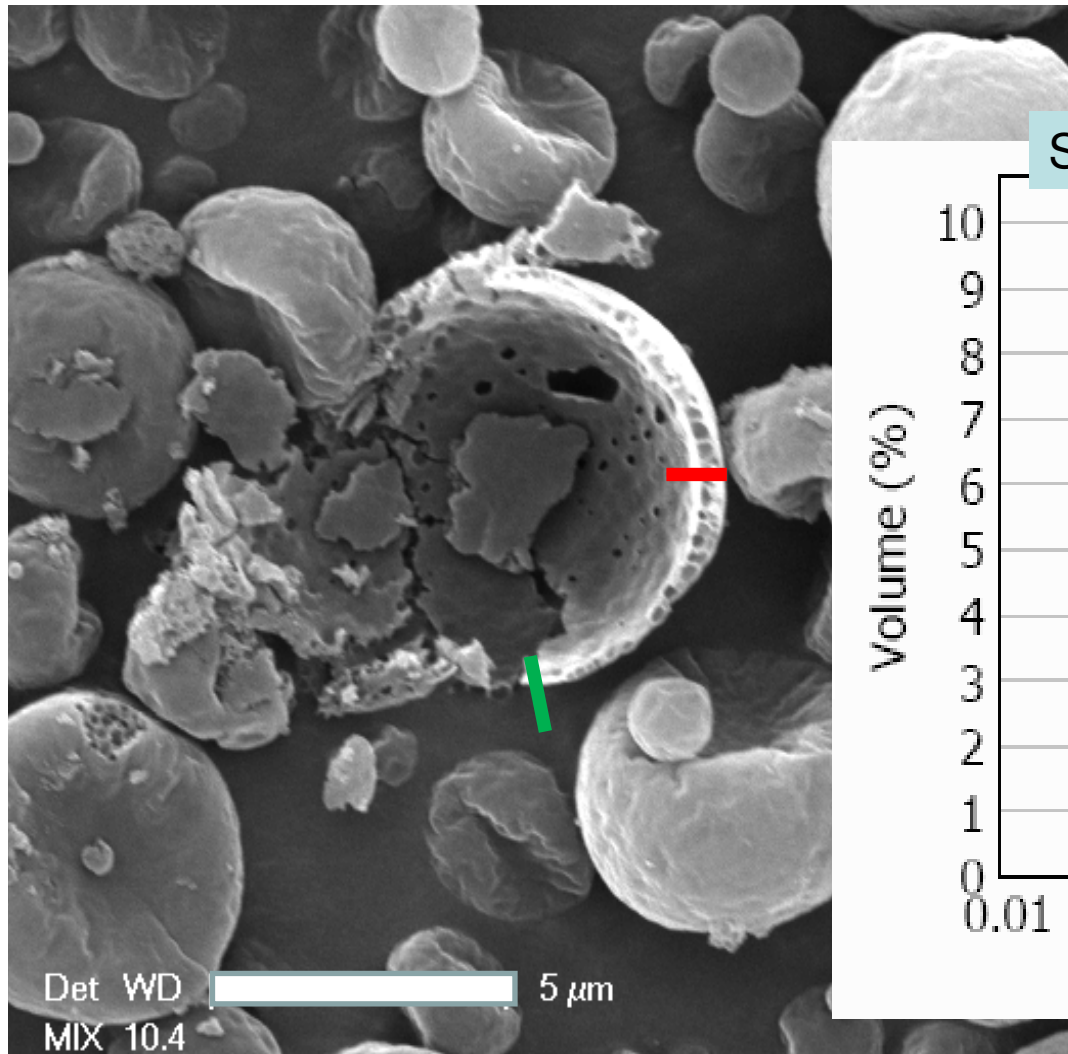


Droplet size range 0.9-3 μm

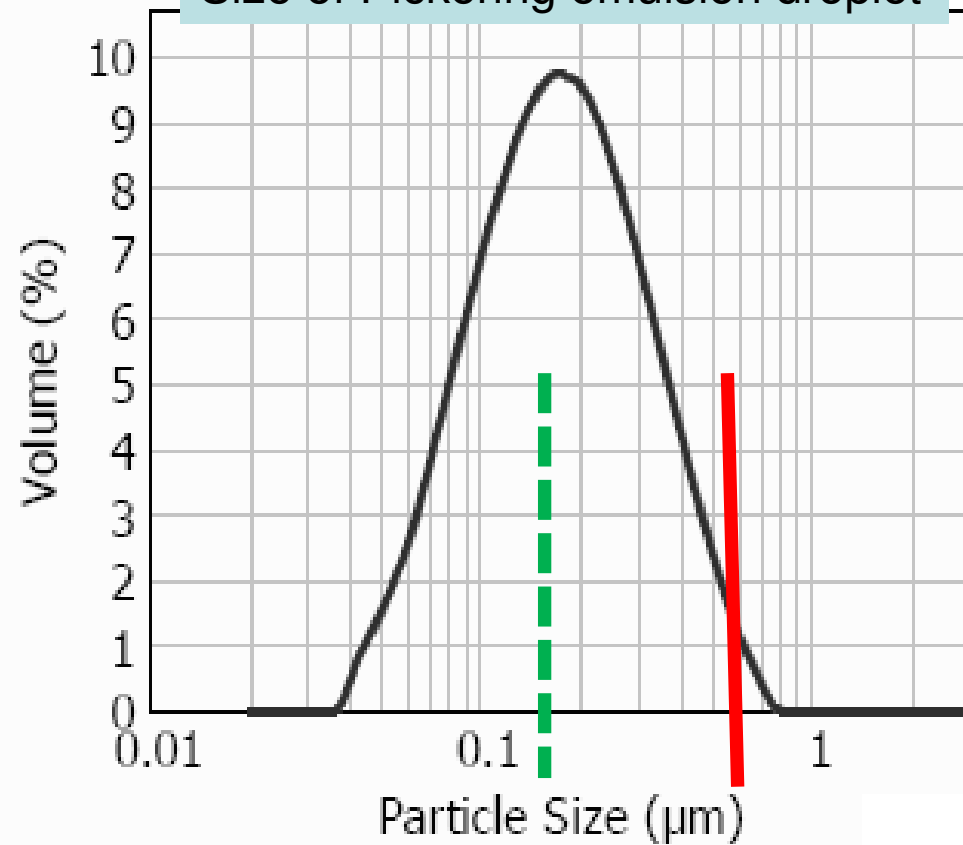


Acc.V Spot Magn Det WD |———| 2 μm  
20.0 KV 3.0 8000x MIX 10.3

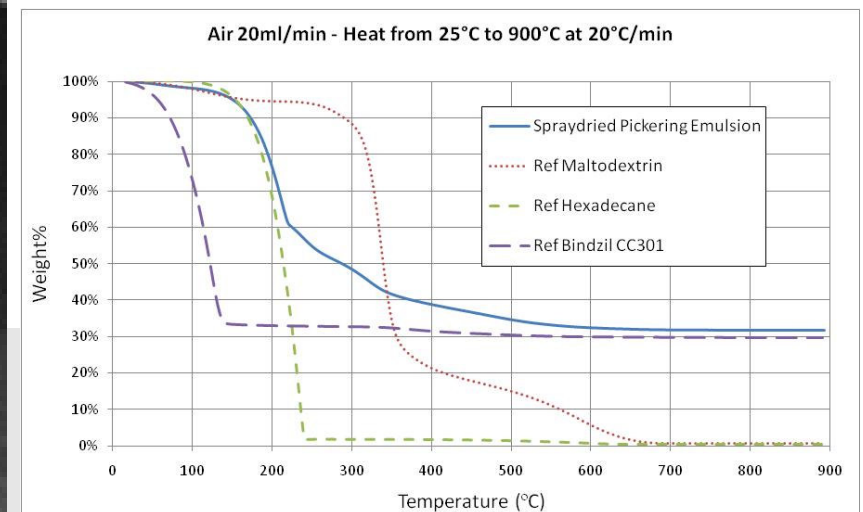
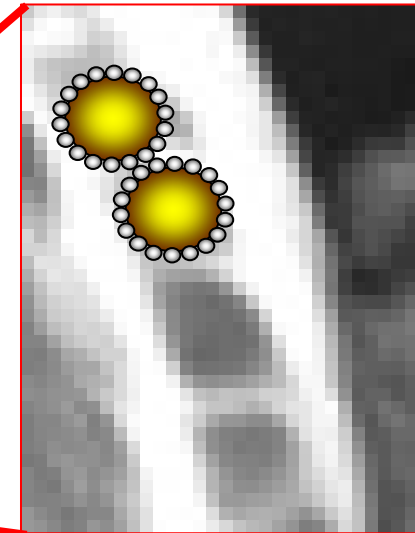
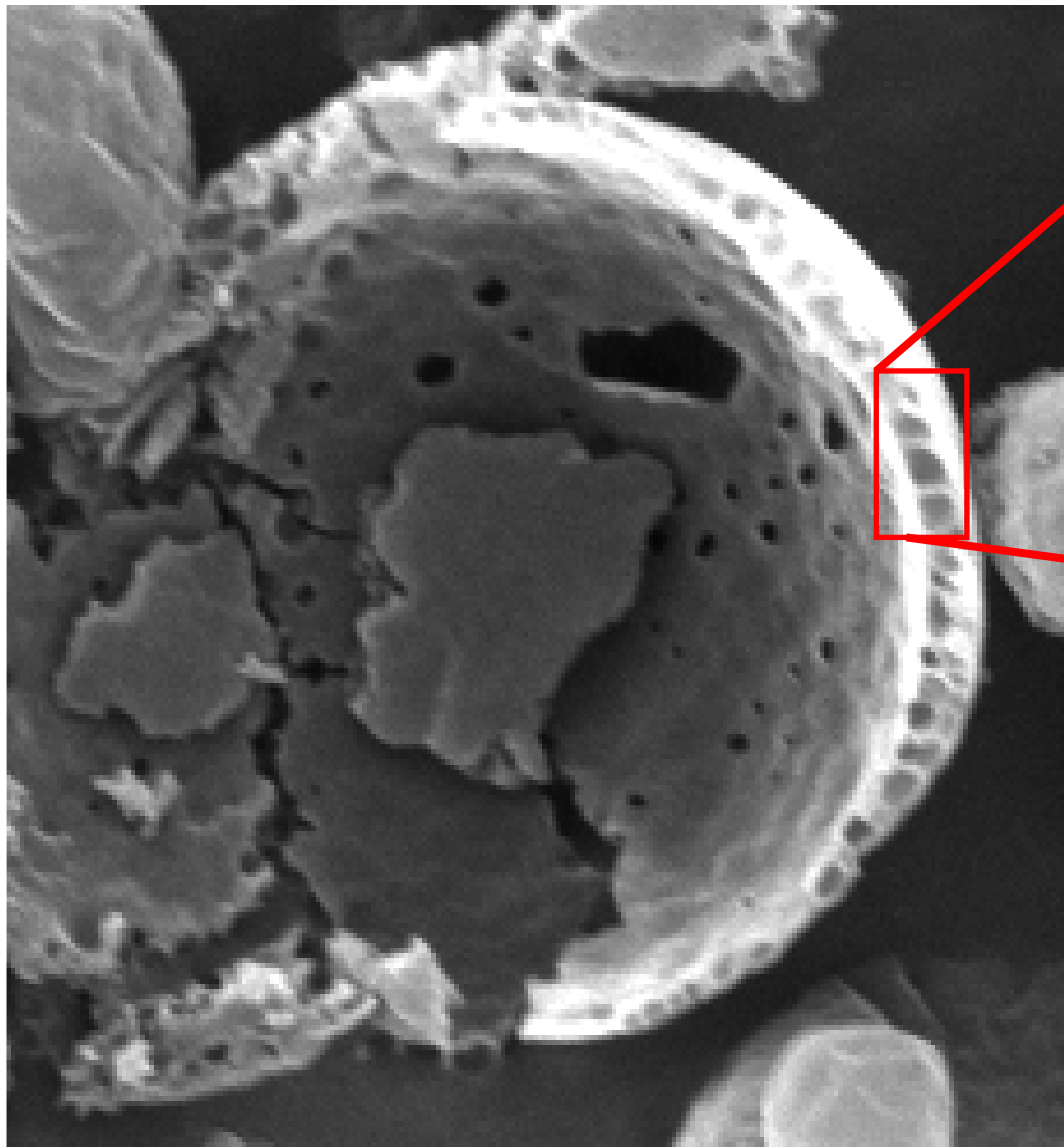
# Spray-dried Pickering Emulsion



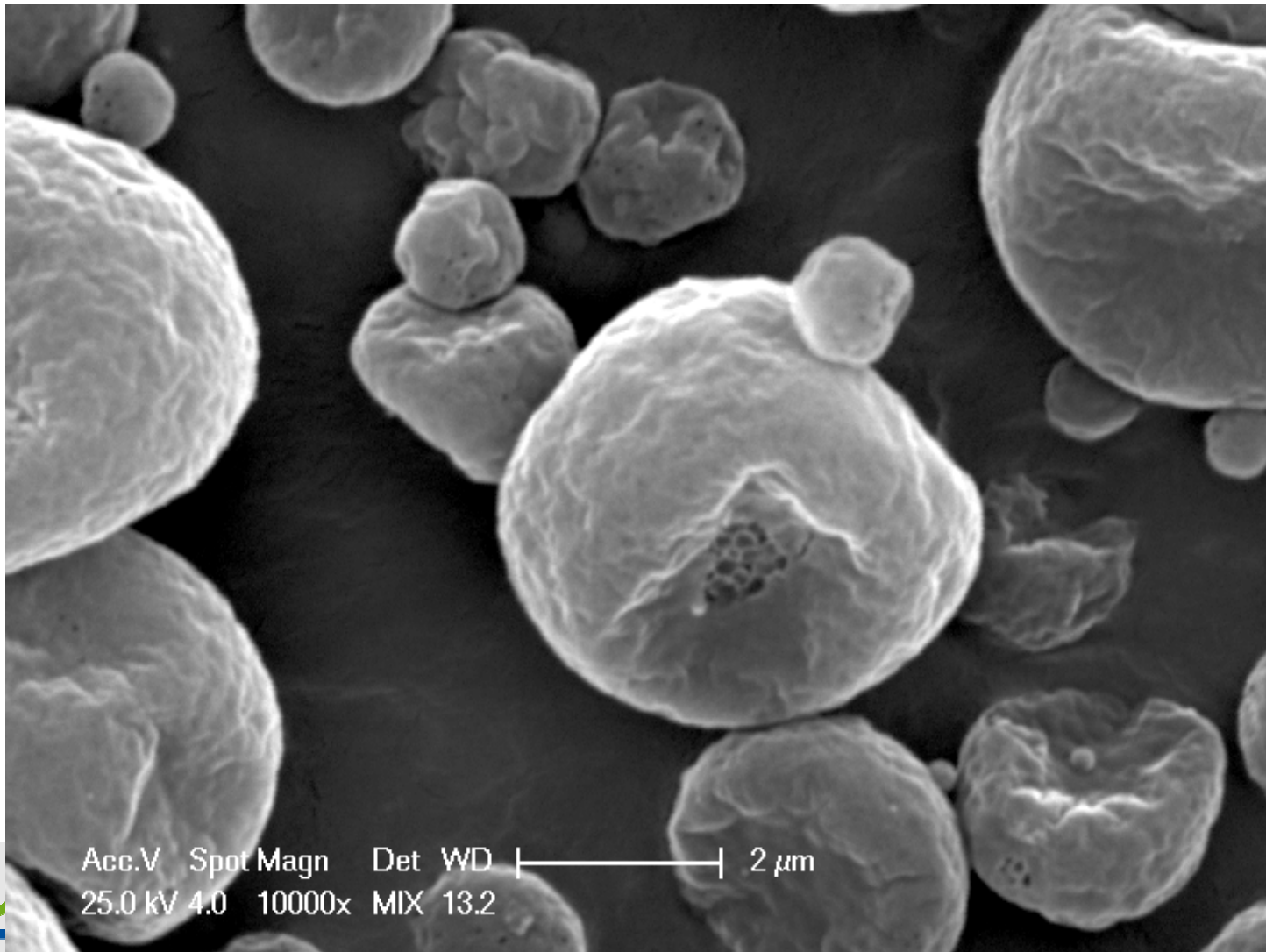
Size of Pickering emulsion droplet



# Environmental SEM of Spray-dried Pickering Emulsion



# Calcined



Acc.V Spot Magn Det WD |—————| 2  $\mu$ m  
25.0 kV 4.0 10000x MIX 13.2

# Summary

Nano-emulsions with

hexadecane, squalene, vegetable oils, silicon oil

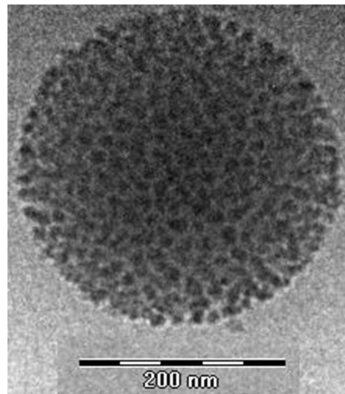
Stable against stress

temperature, pH, surfactants, electrolyte

Colloidosome (microcapsules) formation by spray-drying

Cacination of colloidosome

---



The YKI Pickering emulsion team:

Karin Persson

Anna Fureby

Isabel Mira

Adam Feiler

Eva Sjöström

Johan Andersson

Irena Blute

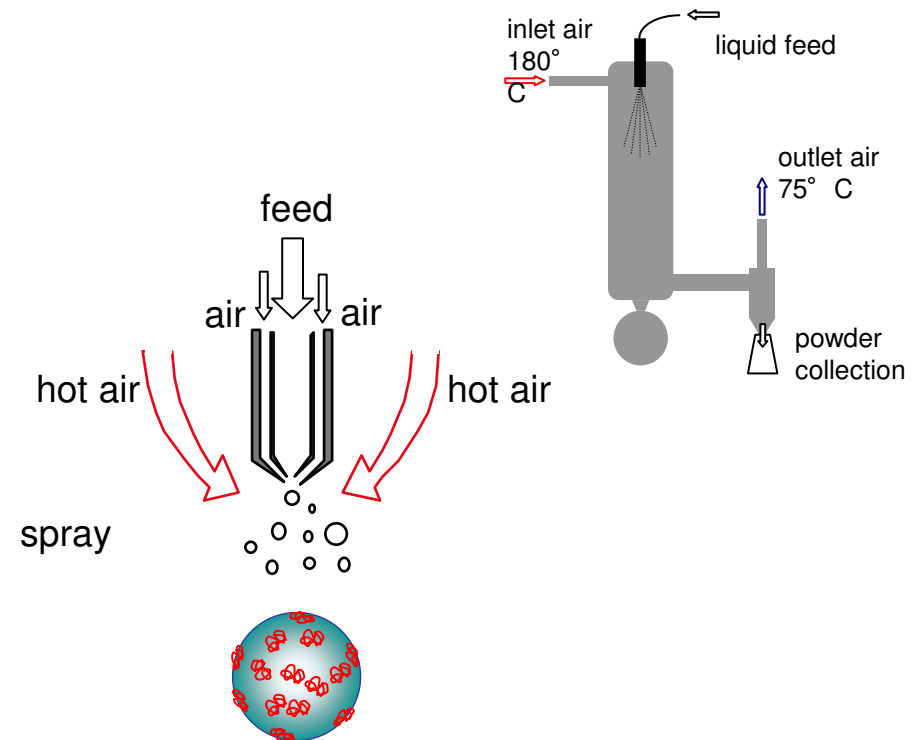
Hans Ringblom

*Thank you for your  
attention!*



# Surface Formation in Spray-drying

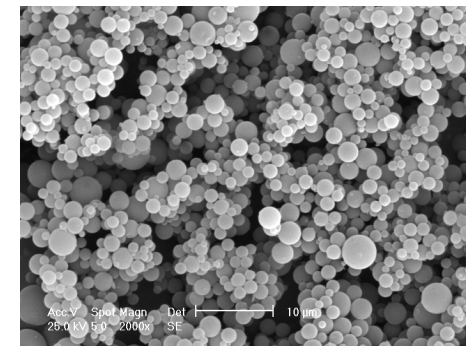
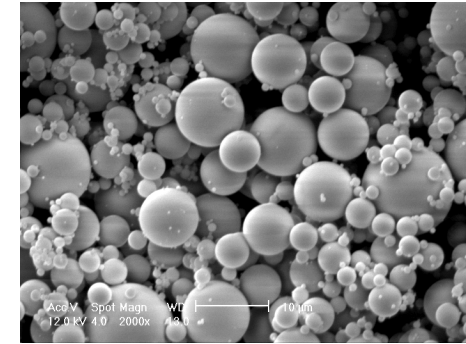
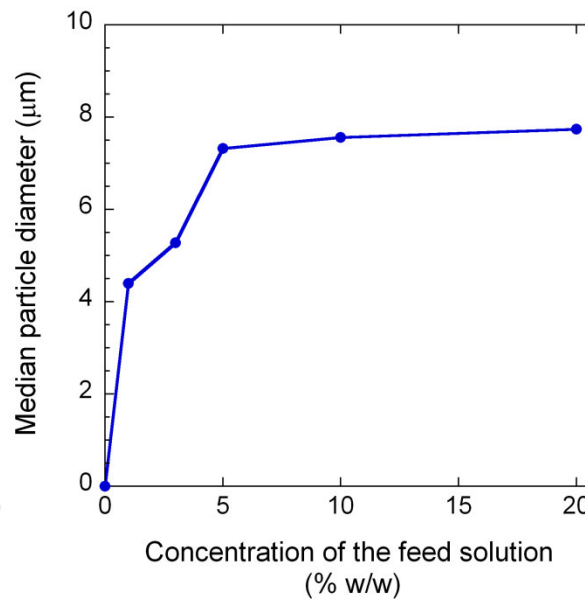
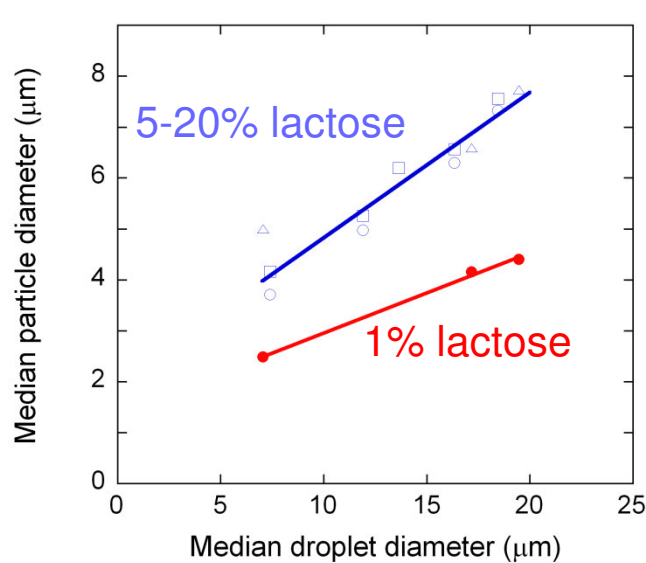
- Surface-active substances adsorb at the air/liquid interface of the spray-droplet
- This surface is conserved during spray-drying
- The surface composition reflects the composition of the spray-droplet surface
- The extent of surface excess of a substance depends on adsorption kinetics and efficiency of the substance and any competing components in the drying liquid



# Spray Drying: Particle Formation

## Control of the particle size

Mainly hollow particles

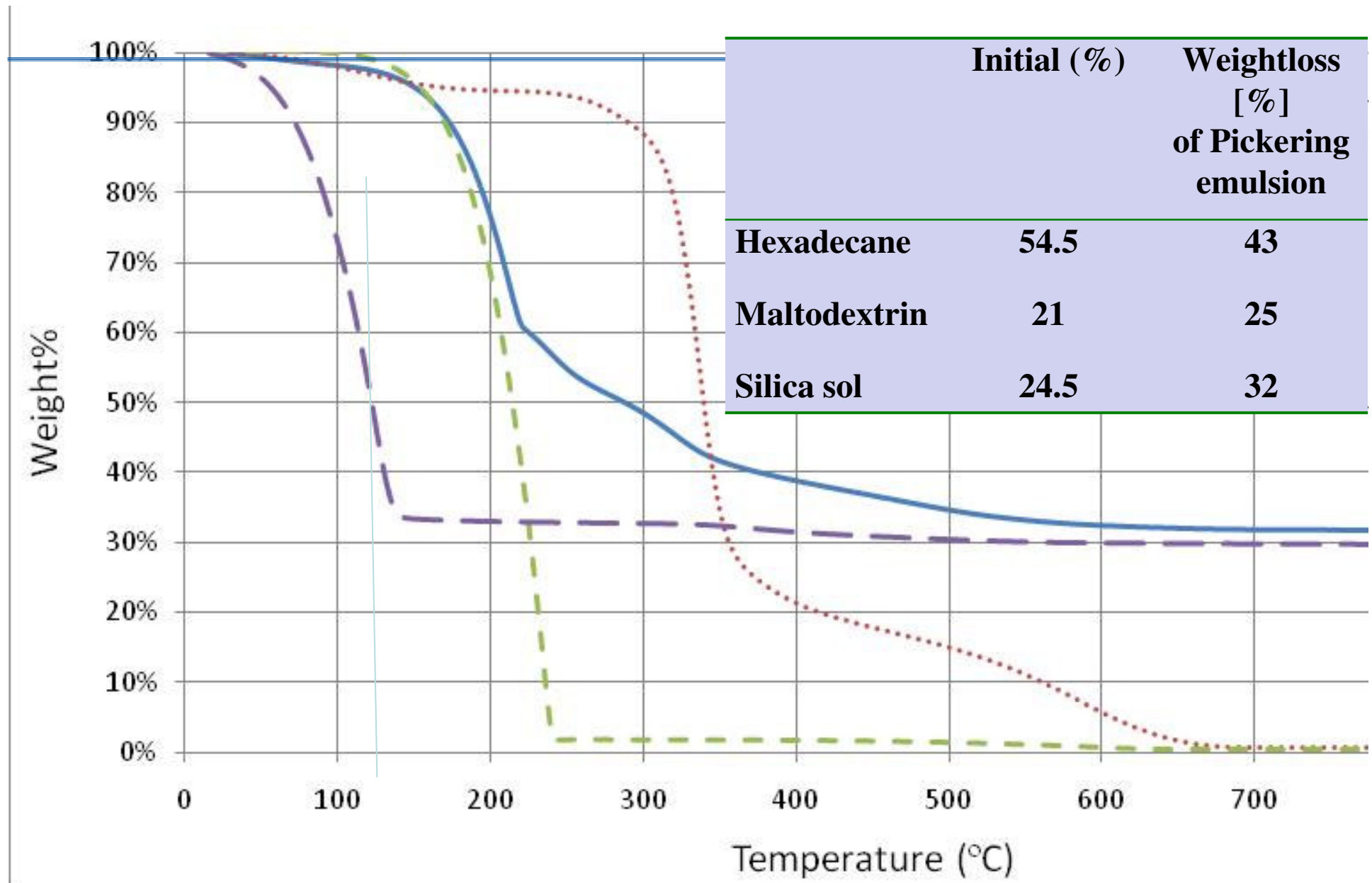


Droplet size  
Change nozzle operation

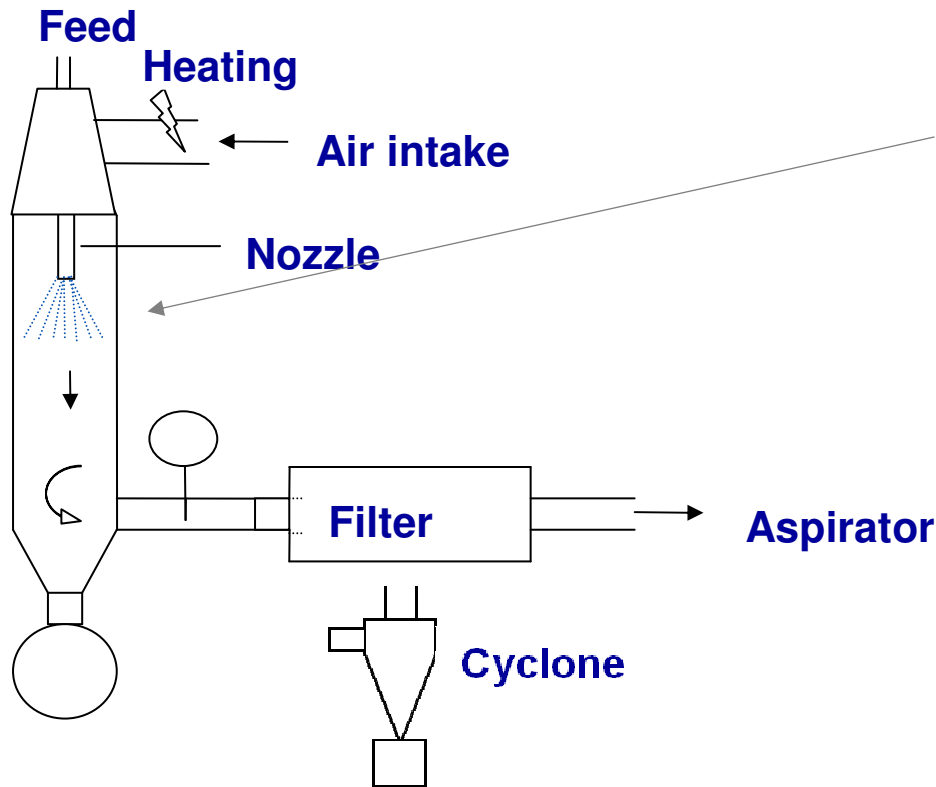
Concentration of  
the spray solution  
(little effect on  $\eta$ )

Reduced surface  
tension

# Content of Emulsion Retained During Spray-drying

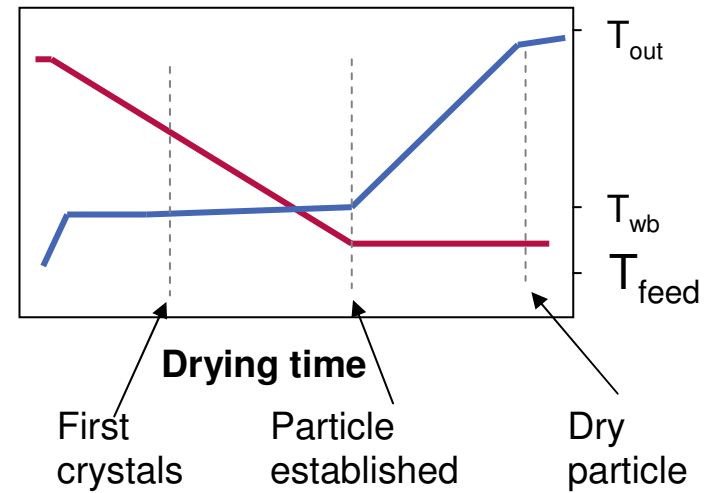


# Spray-drying Process



**Diameter**

**Temperature**



## Droplet life-time during spray drying

4.5 ms (droplet 10  $\mu\text{m}$ , particle 7  $\mu\text{m}$ )  
 21 ms (droplet 50  $\mu\text{m}$ , particle 30  $\mu\text{m}$ )

## Droplet surface temperature:

$T_{(wb)}$ , wet bulb temperature  $\sim 45^\circ\text{C}$   
 (at  $T_{in}=200^\circ\text{C}$  and  $T_{out}=90^\circ\text{C}$ )

**Residence time in lab dryer:  $\sim 1\text{sec}$**