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A Cellular Automata model for the release of corrosion inhibitors from primer coatings

Eugenio Bonetti – CEAS, The University of Manchester, UK / AkzoNobel, Malmö, SE Flor R. Siperstein – CEAS, The University of Manchester, UK Peter Visser – AkzoNobel, Sassenheim, NL Simon R Gibbon – AkzoNobel , Felling, UK



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Inhibitors release from primers

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Corrosion inhibitors leach out when the primer is exposed to water and protect the metal

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Develop a model of the release of inhibitors to investigate the relationship between coating formulation and leaching behaviour





Experimental evidence

- Pigment particles form clusters
- Clusters dissolution leaves interconnected cavities
- Pigment volume concentration and particle size distribution determine the microstructure
- Inhibitor properties (e.g. solubility) affect the release rate



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[1] Trueman et al., *Corr. Science.*, 75 (2013) 376-385 [2] Emad et al., *Prog. Org. Coat.*, 134 (2019) 360-372



Model implementation

Step 1 – Microstructure generation



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Step 2 – Release simulation







Automaton states and rules

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Microstructure generation

- Pigment volume concentration
- Particle size distribution



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Clusters formation







SUSTICOAT

Simulation approach: Cellular Automata model

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Discrete approach

Transition rules

Locality







Release simulation

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Simulation setup



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Microstructure analysis

Average number of particle clusters



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Average size of particle clusters





Clusters percolation



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The leachable fraction of inhibitor shows a percolating behaviour as a function of the PVC





Parametric study of the effect of PVC



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Increasing PVC results in increased release due to increased particle connectivity







Parametric study of the effect of PVC

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Similar trends, but different amount released



Parametric study of the effect of particle size

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Short time release: controlled by pigment volume fraction (same for all systems)

Long time release: controlled by internal coating structure (determined by particle size)



Parametric study of the effect of particle size





Similar trend, similar amount released



Parametric study of the effect of inhibitor solubility



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140 Low solubility leads to 120 Mass released per unit area / μg cm⁻² slow initial release **PVC = 30%** (critical for protection) 100 Solubility 80 $S = 0.13 \text{ g/cm}^3$ High solubility leads to 60 $S = 0.013 \text{ g/cm}^3$ fast depletion $S = 0.0013 \text{ g/cm}^3$ 40 20 0 5 10 15 20 25 0 30 Time / h





Parametric study of the effect of inhibitor solubility



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Conclusions

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- The model qualitatively reproduces the experimental correlations between leaching behaviour and formulation parameters (e.g. PVC, size, solubility), allowing parametric investigations of the effect of the formulation
- Quantitative comparison with experimental data is limited, suggesting that other factors not included in the model may play a role in the release process
- Better understanding of the phenomena is needed to direct the modelling work and identify what is missing in the model
- Quantitative characterization of coating microstructure is needed to validate the virtual coating model



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