



The University of Manchester



H2020-MSCA-ITN-2016/721451



PARTICLE LEACHING FROM POLYMERIC COATINGS A Combined Experimental and Simulation Study

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FORMULA X
Manchester, UK

Introduction

Active corrosion protection by organic coatings in Aerospace

Corrosion in modern world

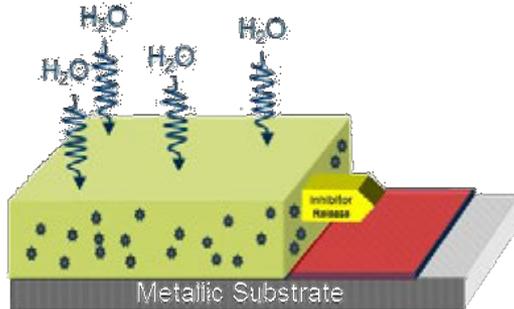
- Corrosion cost: 2.5 trillion US\$ - 4% percent of the global Gross Domestic Product (GDP)
- 20-35% of this loss could have been saved by implementing better corrosion preventing practices. German, T. & Section, N. Corrosion news. 140 146 (2018).



In the aerospace industry corrosion impacts

- Cost
- Aircraft availability
- **Safety**
- **Social**

Organic coatings provide corrosion protection



Key steps for corrosion inhibition via leaching

1. Water uptake
2. Dissolution of active pigment
3. Transport and delivery of active species through the polymeric matrix to de defect area
4. Fast, effective, and irreversible passivation

O Gharbi et al. *Npj Materials Degradation*, 2, 12 (2018)

October 1992, The Netherlands:

Fatigue corrosion cracking

<https://aviation-safety.net/database/record.php?id=19921004-2>



April 1988, Hawaii: Fatigue corrosion cracking

https://faculty.up.edu/lulay/me401/aloha_flight_243_a_new_direction.pdf



Introduction

Active corrosion protection by organic coatings in Aerospace

Challenge

Develop coatings that are:

- Environmentally friendly
- Sustainable
- Efficient
- Cost effective

Approach

- Replacing toxic chemicals (chromates)
- Use renewable materials
- Understand and optimise performance
- Minimise use of expensive materials

UNDERSTAND LEACHING OF INHIBITORS FROM ORGANIC COATINGS

UNDERSTAND THE RELATIONSHIP BETWEEN LEACHING AND PERFORMANCE

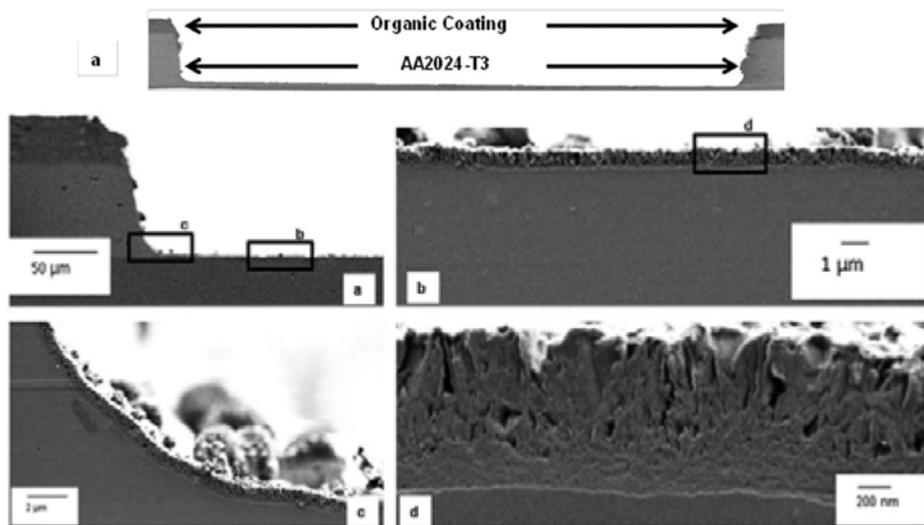
Introduction

Lithium leaching technology protection concept



Provide **fast**, **effective**, and **irreversible** corrosion inhibition

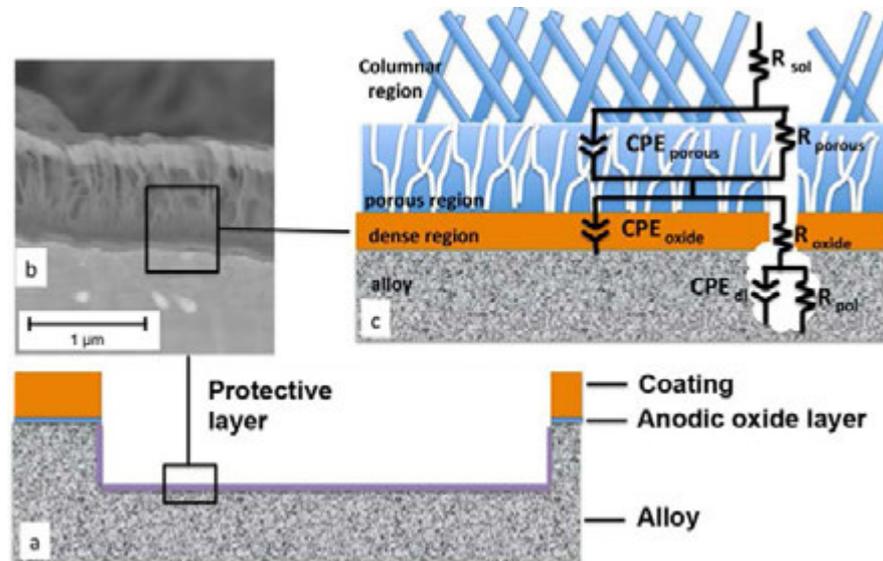
1. Leaching of lithium ions
2. Lithium ion transport to defect area
3. Formation of a protective layer on the aluminium substrate



Visser et al. *Faraday Discuss.* **180**, 511-526 (2015)

Fitting with physical model

- Oxide layer provides corrosion protective properties



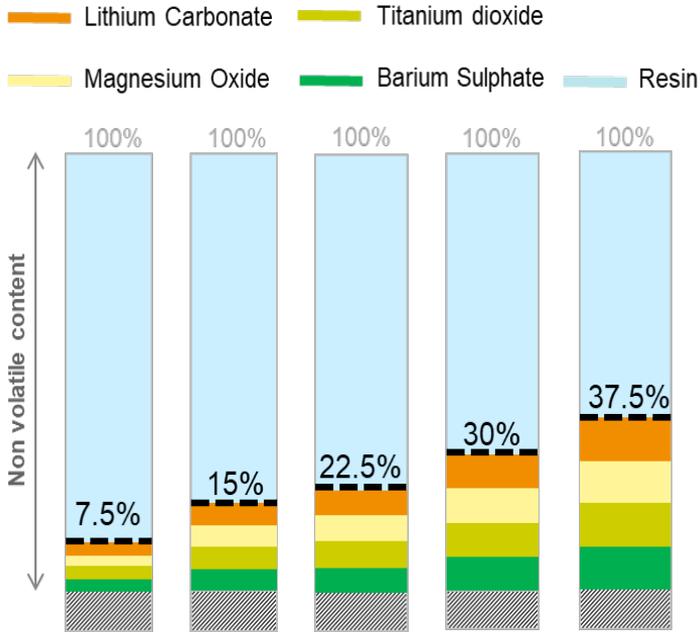
Visser et al. *Journal of The Electrochemical Society*, **164** (7) C396-C406 (2017)

The effect of the PVC

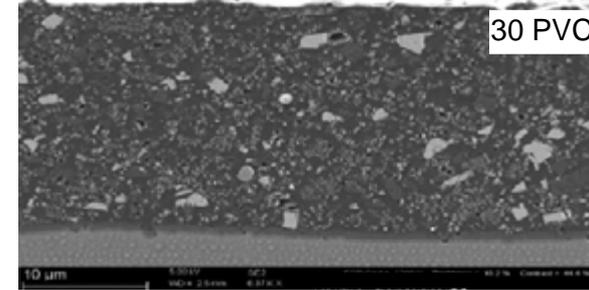
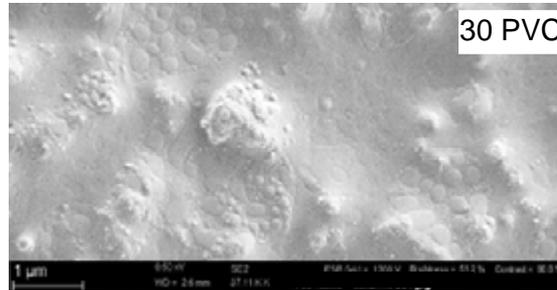
Systems of study and microstructure analysis

Epoxy model system with PVCs between 7.5 – 37.5%

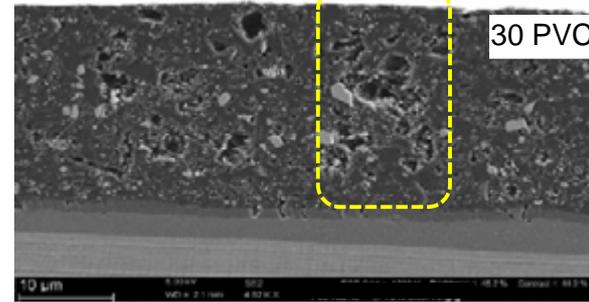
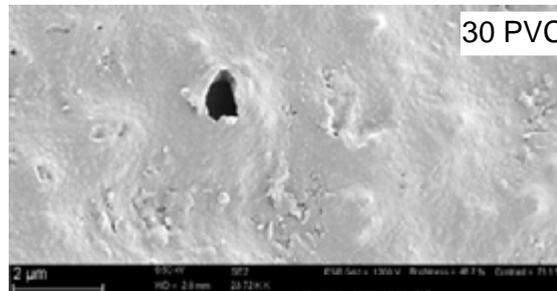
Pigment ratios constant



Before exposure



After 2 weeks exposure to 3.5% NaCl solution

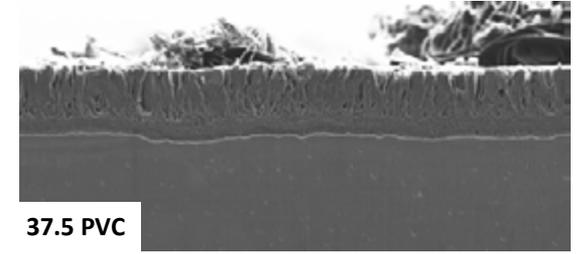
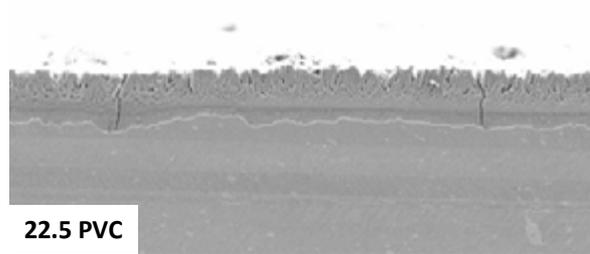
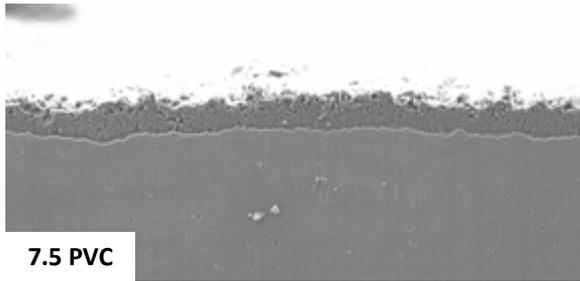
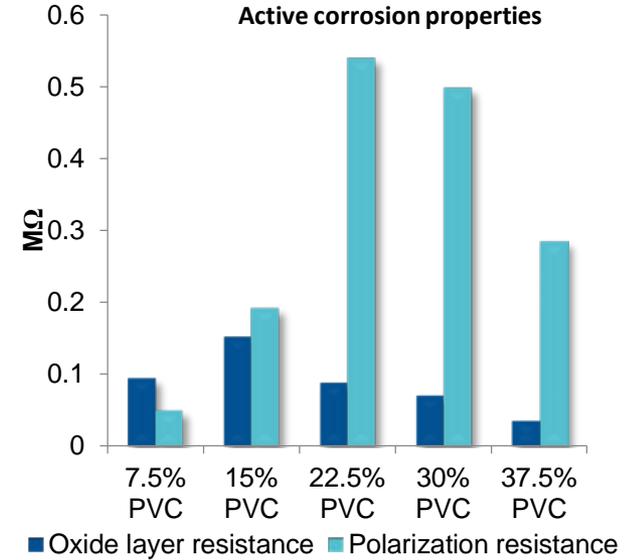
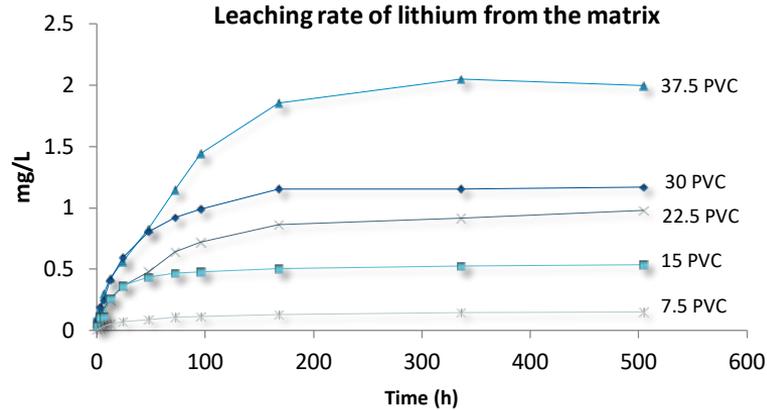
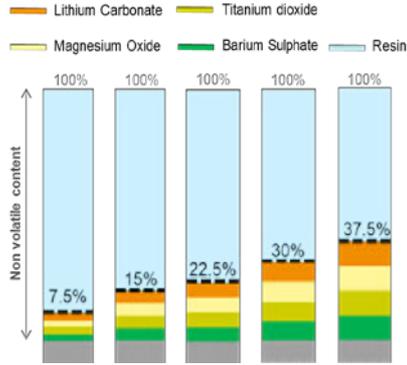


The effect of the PVC

Lithium leaching and corrosion protection properties

Epoxy model system with PVCs between 7.5 – 37.5%

Pigment ratios constant



The effect of the PVC

Higher pigment volume concentration result in higher leaching rates and faster exhaustion of the system

- Larger network of interconnected pigment clusters

Increase of pigment volume concentration decrease the barrier properties of the coating

- Formation of defects through the polymeric matrix which enhance water permeation

Slower dissolution rate leads to higher R_{oxid} and R_{pol} – Better active corrosion protection

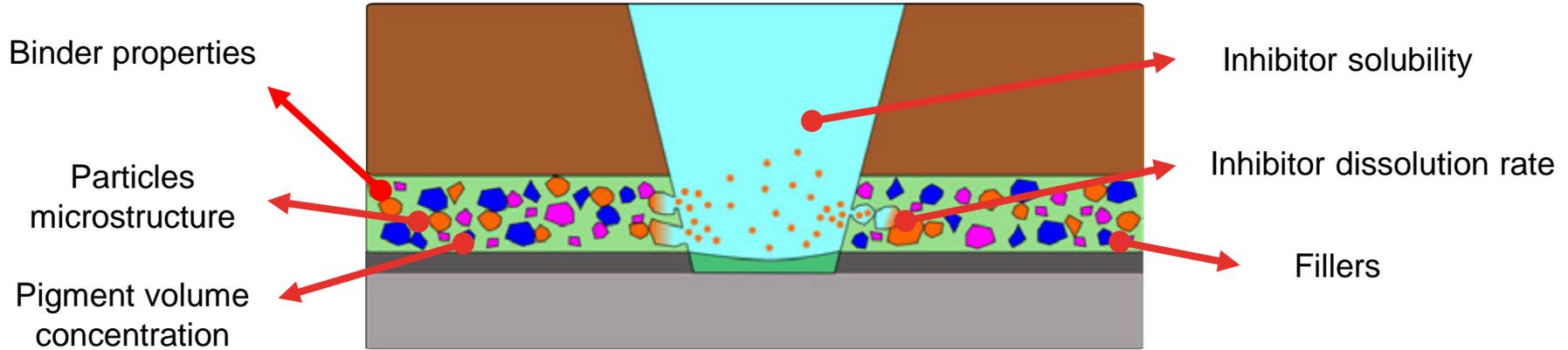
Can we control the rate of release?

Can we control the amount released?



Develop a model that describes the leaching of inhibitors with sufficient accuracy to explore the formulation space and inform what are the desirable features or properties of the coating and inhibitor

Release of corrosion inhibitors

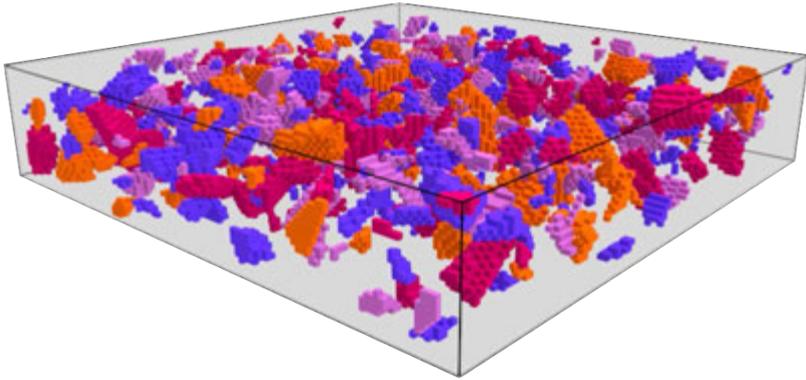


Cellular Automata Model is a good compromise to model the release of inhibitors:

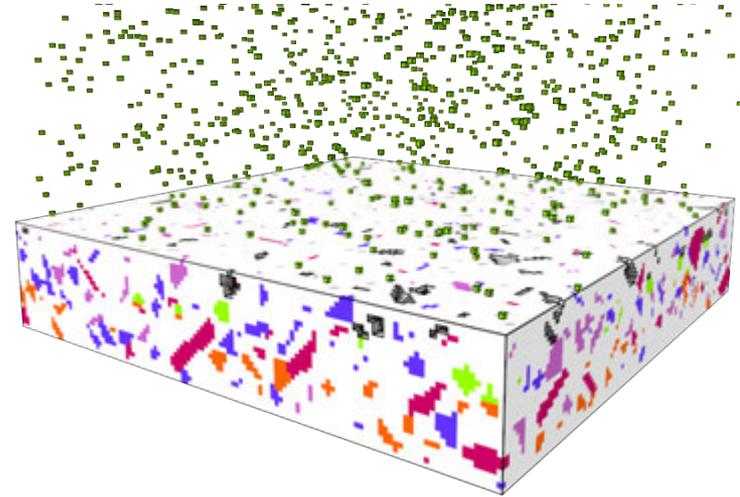
- ✓ Can capture different physical phenomena
- ✓ Can be related to real physical properties of the system
- ✓ Computationally efficient

Approach

Microstructure generation

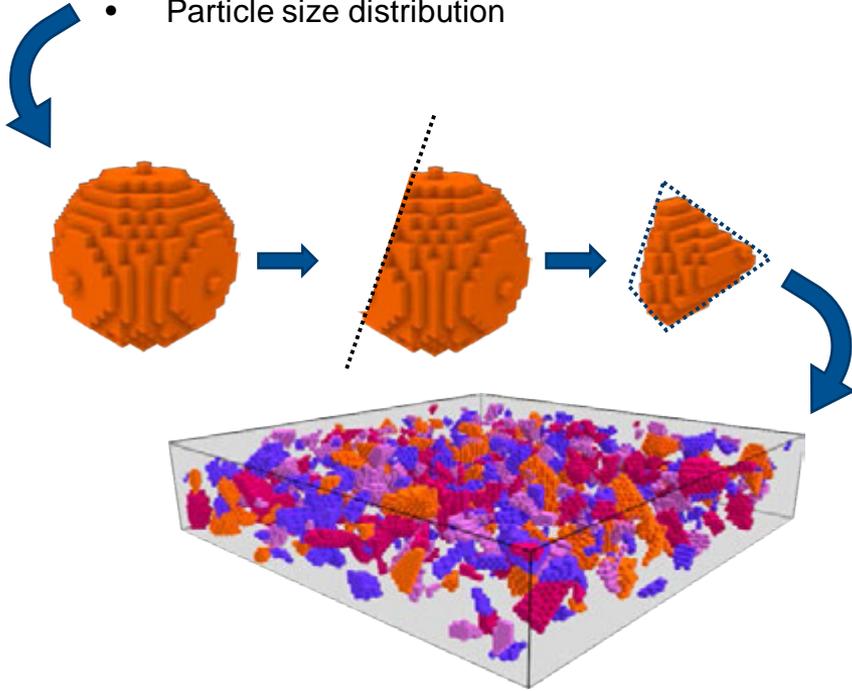


Release simulation

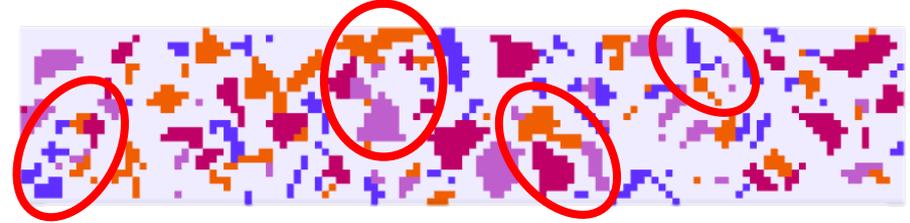


Microstructure generation

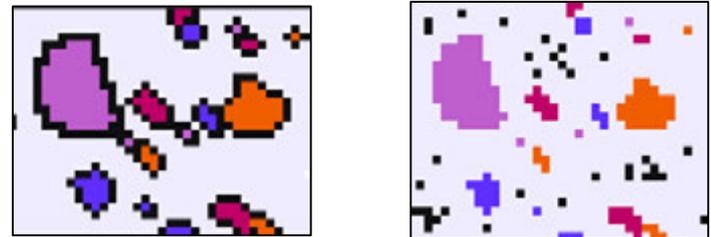
- Pigment volume concentration
- Particle size distribution



Clusters formation

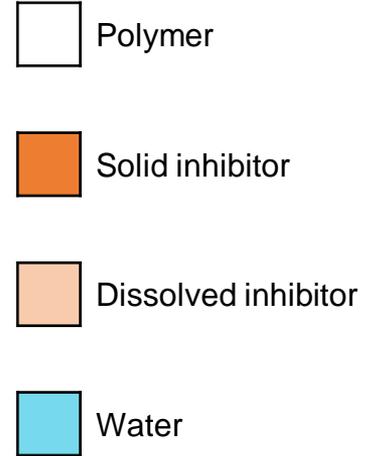
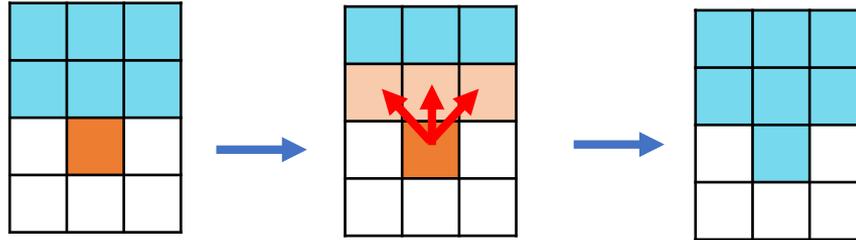


Permeable heterogeneities



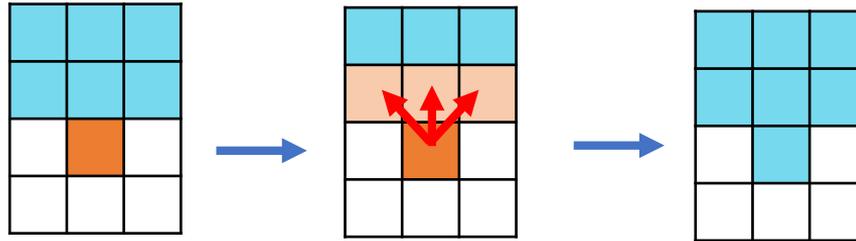
Release simulation

Dissolution

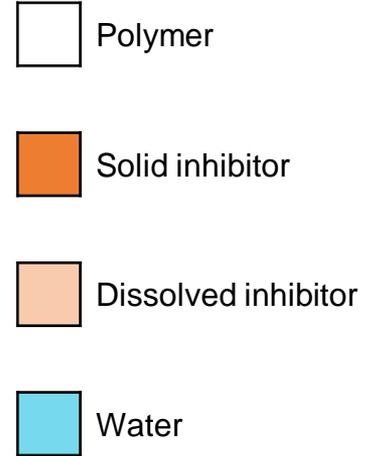
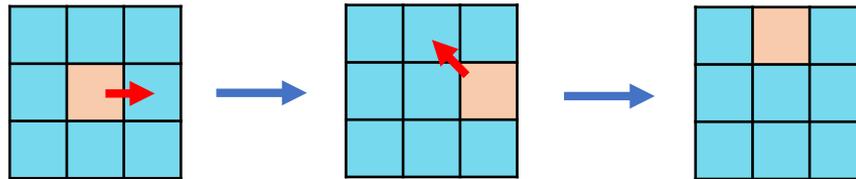


Release simulation

Dissolution

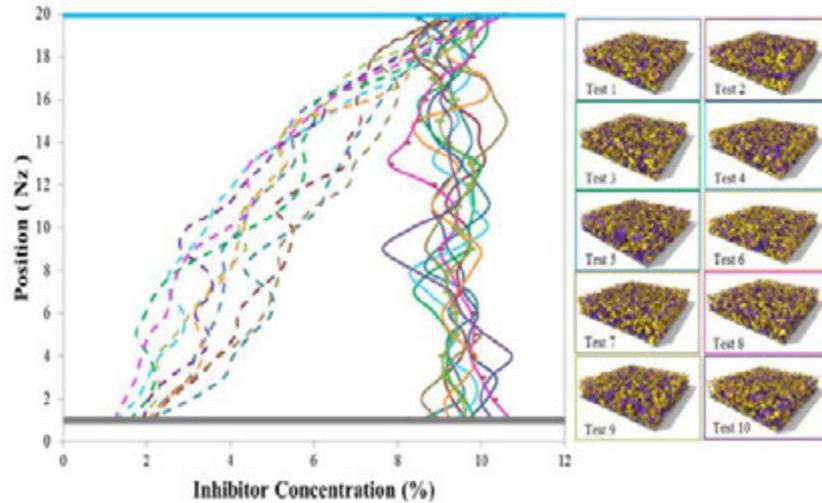


Diffusion

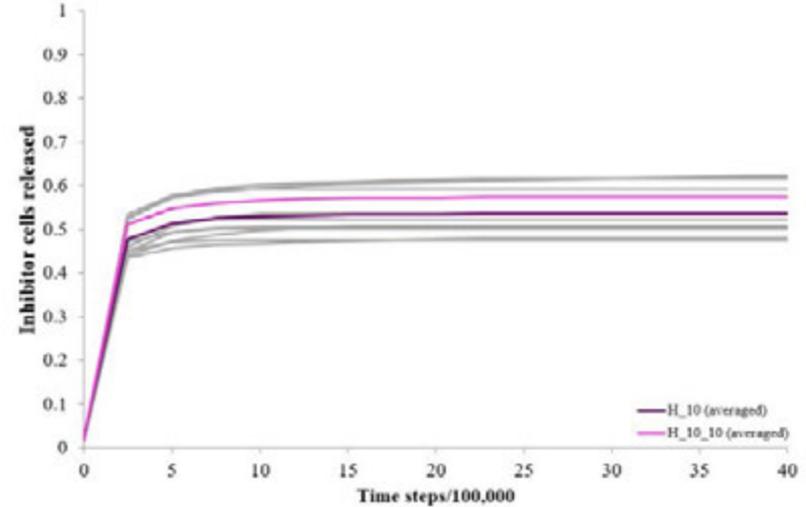


Results

Total and connected inhibitor concentration

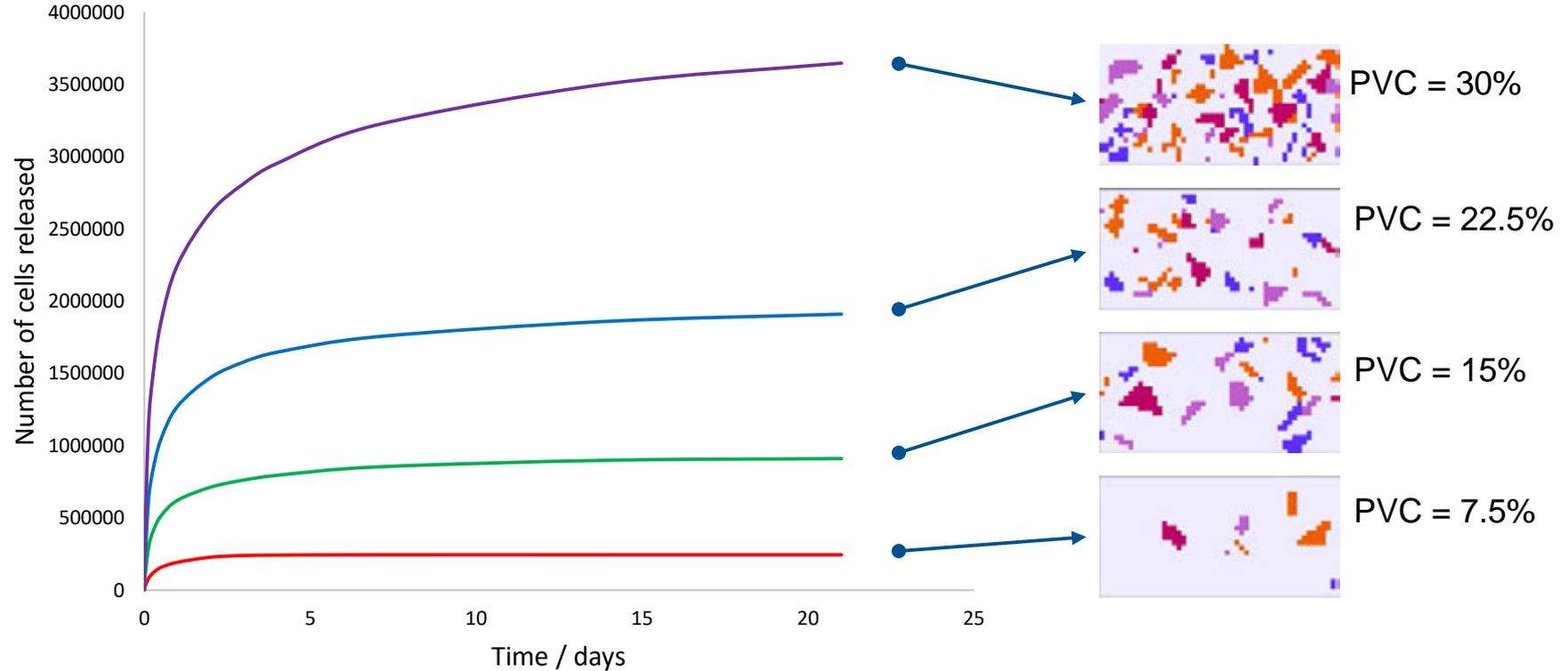


Release curves

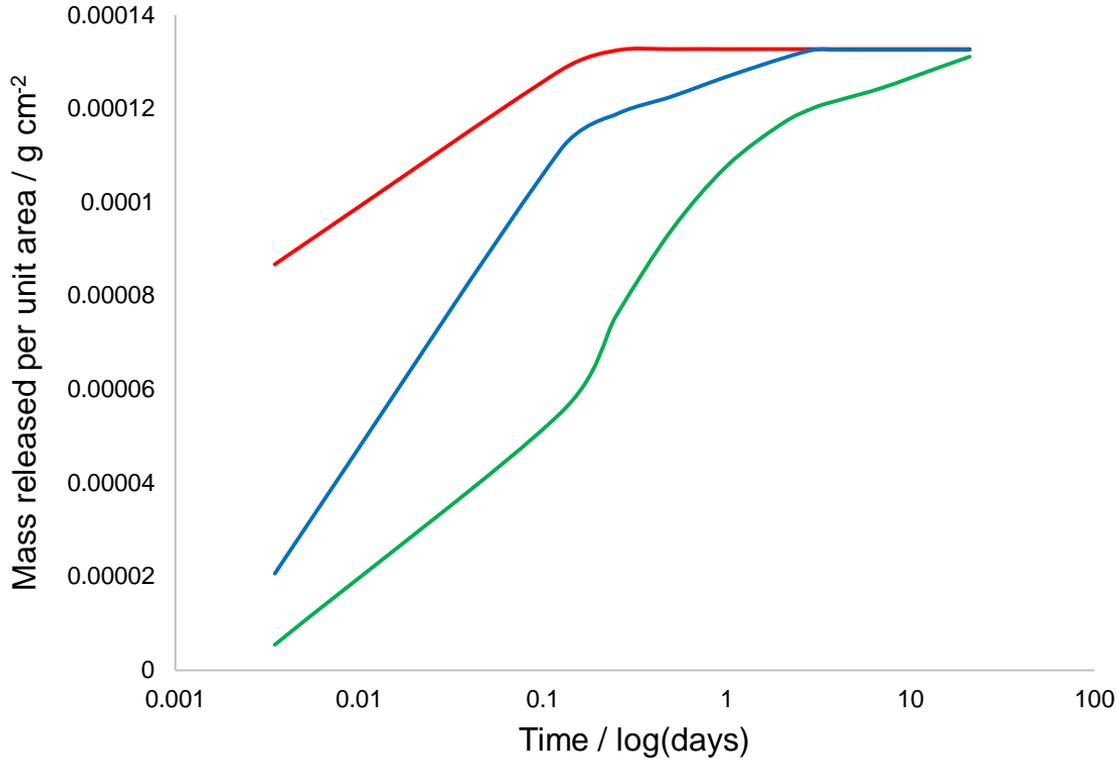


Random distribution of inhibitor particles results in unique connectivity profiles.
Averages are needed over multiple configurations to represent real systems

Effect of PVC



Effect of solubility



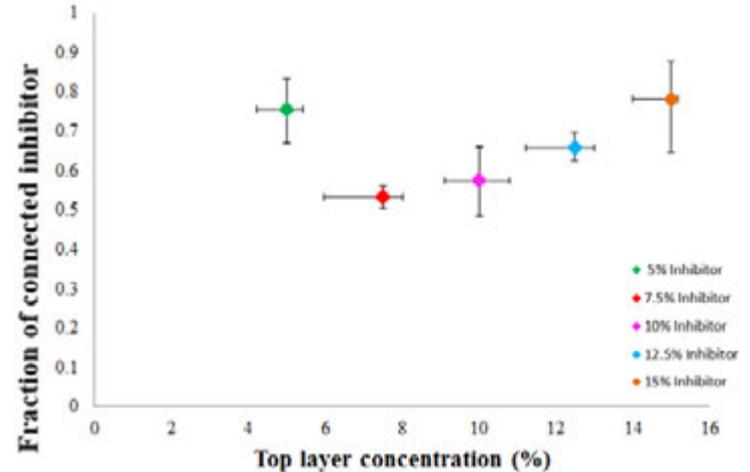
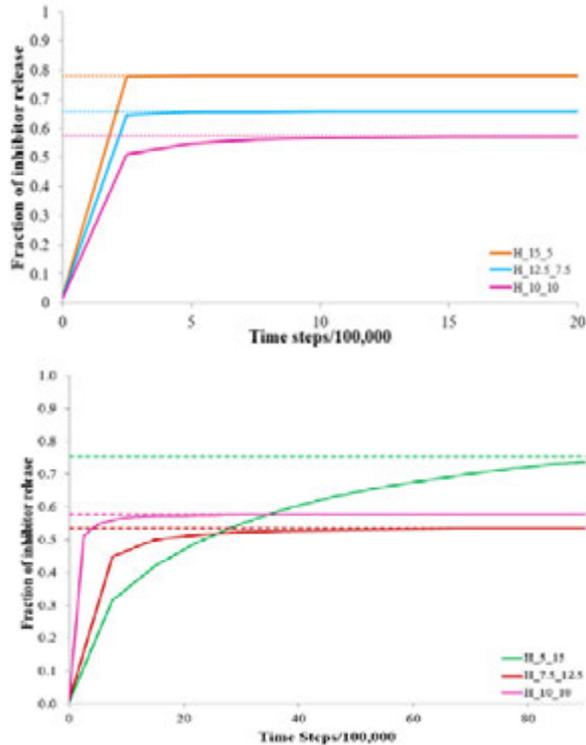
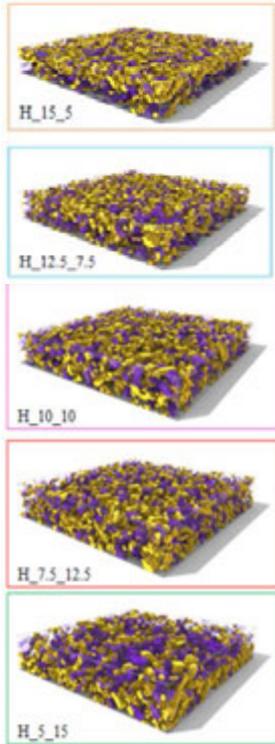
Solubility

$$S = 0.13 \text{ g/cm}^3$$

$$S = 0.013 \text{ g/cm}^3$$

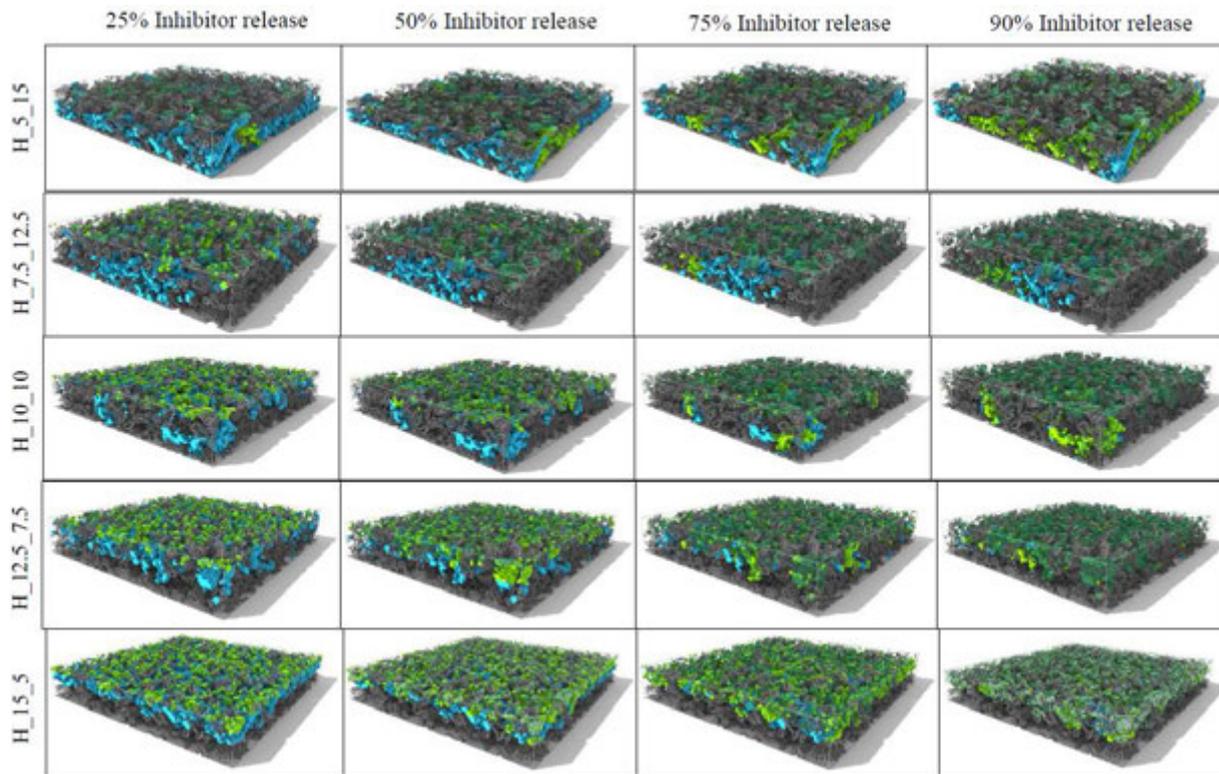
$$S = 0.0013 \text{ g/cm}^3$$

Effect of particle distribution

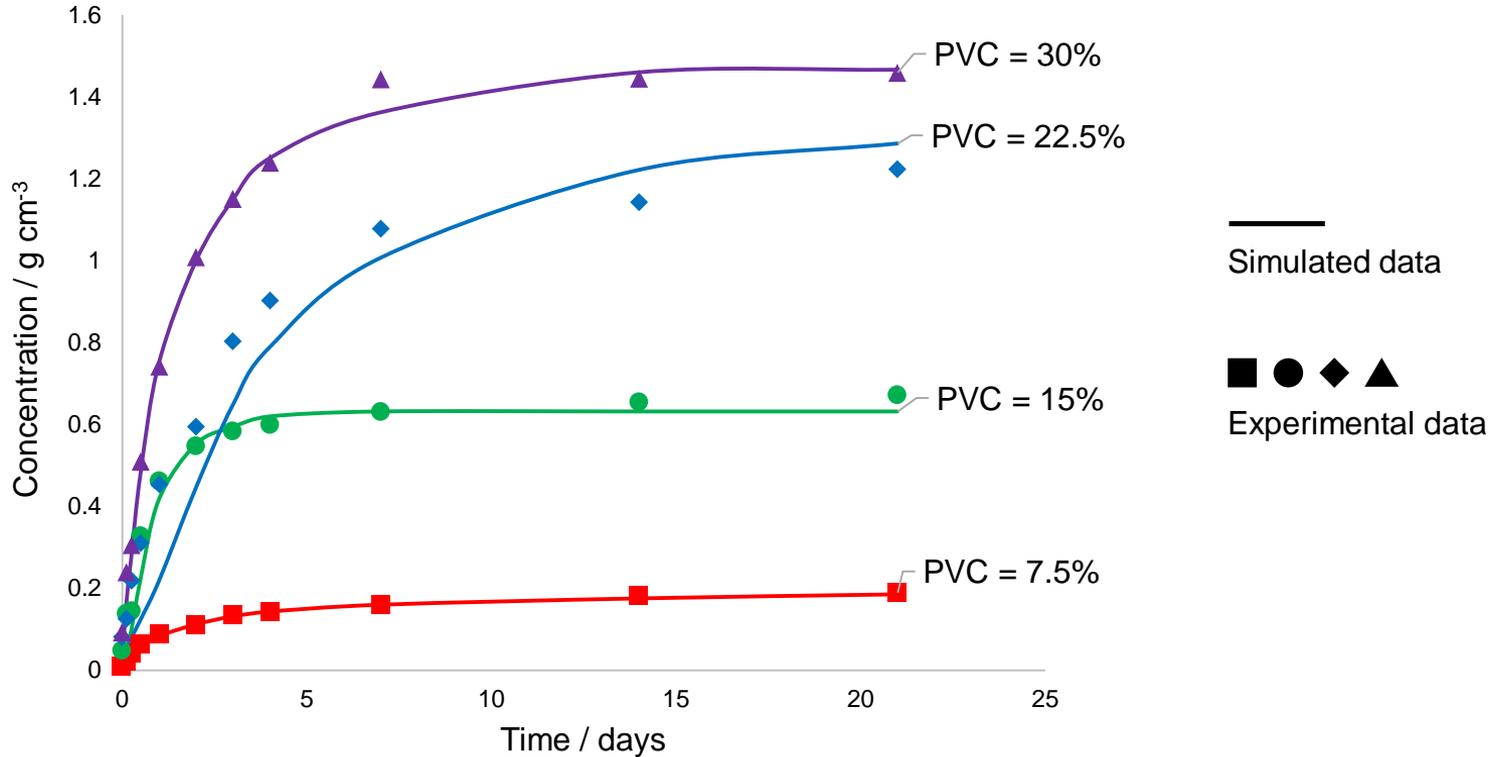


Spatial heterogeneity can affect the connectivity between inhibitor particles

Effect of particle distribution



Comparison with experimental data



Conclusions

- The release of corrosion inhibitors from primer coatings can be modelled using the **CA approach**
- Models can show the influence of **microstructure** and **pigment properties** on the release, enabling **control** on the factors that affect the process
- Simulations can provide important insight on the structure-property relationship in complex coatings to enable **optimal formulation design**

Acknowledgments

Dr. Yanwen Liu

Ms Maria Georgiades

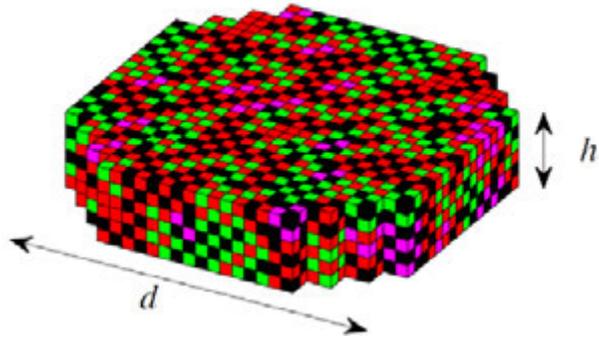
Assistance given by the IT Services and the use of the Computational Shared Facility at the University of Manchester.

University of Manchester – AkzoNobel Corrosion Protection Partnership

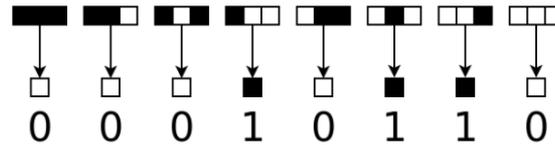
This project has received funding from the European Union's Horizon 2020 research and innovation programme under the Marie Skłodowska-Curie grant agreement No. 721451

Cellular Automata model

Discrete approach



Transition rules



Locality

