



The processing route to unique formulations

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EFFECT OF SUGARS ON MIXED BIOPOLYMER SYSTEMS

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Background

□ Mixed biopolymers are of direct industrial significance.

- Proteins, polysaccharides and sugars are major nutritive, texture and taste imparting components in dairy emulsions.
- Stability, taste and texture are responsible for successful marketing and increased food consumption.
- Structure of these foods is built by intermolecular interactions between the components.









Aim and Objectives

Explore the process of texture formation in sodium caseinate-galactomannan-sugar model systems.

- <u>Part 1</u>: Liquid-liquid phase separation \Rightarrow w/w emulsions
- <u>Part 2</u>: Deformation of the emulsions in flow \implies microstructure at quiescent conditions and in flow
- <u>Part 3</u>: Interfacial properties (effect of composition) > new texture

Experimental work

Cosolubility of sodium caseinate-galactomannansugar systems studied by phase-volume ratio method

The effect of composition on the microstructure at rest and under shear visualised by microscopy and optical rheology

The interfacial properties of the systems calculated by drop retraction method







Part 1: Phase behaviour



- binodal
- tie lines
- characteristic points

-galactomannan type -sugar concentration -sugar type

Sodium caseinate concentration [wt%]

2D composition-composition isothermal phase diagram

Part 1: Phase behaviour

Effect of galactomannan type

Effect of sugar concentration

Effect of sugar type



Conclusions (Part 1)

Highly branched galactomannans improve the cosolubility of the biopolymers.

Sugar content below the critical concentration improves the cosolubility.

Increasing sugar concentration reduces the cosolubility.

At 15% of sugar, fructose and trehalose improve the cosolubility most significantly.

Part 2: Effect of volume fraction on microstructure



Polysaccharide-rich phase Protein-rich phase NaCAS-LBG (no sugar)

Part 2: Effect of volume fraction on microstructure

Effect of sugar concentration on the evolution of the microstructure in flow



Conclusions (Part 2)

Under quiescent conditions:

- Phase sense depends on the volume fractions of the equilibrium phases.
- Phase inversion occurs at about equal volumes.

Under shear:

For low sugar, microstructure depends on the volume fraction and applied shear.

□ For high sugar, microstructure depends on the volume fraction only.

Part 3: Interfacial properties

Drop retraction method



- 1. Droplets are deformed under shear and when this is removed their retraction is recorded.
- 2. Deformation parameter **D** vs. time.

$$D = \frac{L - W}{L + W}$$

- 3. Characteristic time **τ**.
- 4. Viscosity ratio λ .



5. Interfacial tension σ

$$\sigma = f(\eta_c, \lambda, R_0, \tau)$$

Part 3: Interfacial properties



NaCAS-LBG (no sugar)

Part 3: Interfacial properties



The above points represent systems of 4% NaCAS-0.2% GM/LBG.

Conclusions (Part 3)

Power law relationship exists between the interfacial tension and the measured parameters.

The interfacial tension in a system of a constant biopolymer concentration depends on the type of galactomannan, sugar concentration/type.

Summary of findings

- Sugars below critical concentration improve mixing behaviour of the biopolymers and reduce interfacial tension compared with the systems without sugar.
- □ A strong link between the microstructure and rheology of the systems is observed.
- Shear-induced phase-inversion observed in the low sugar environment is absent in the high sugar environment.

Related literature

- Frith W.J. (2010) Mixed biopolymer aqueous solutions-phase behaviour and rheology. Advances in Colloid and Interface Science 161:48-60
- Guido S., Greco F., Villone M. (1999). Journal of Colloid and Interface Science 219:298-309.
- Polyakov V. I., Grinberg V. Ya., Antonov Yu. A., & Tolstoguzov V. B. (1979). Polymer Bulletin, 1, 593–597.
- Schorsch C., Clark A.H., Jones M.G., and Norton I.T. (1999). Colloids and Surfaces B: Biointerfaces, 12, 317-329.
- Semenova, M.G., Antipova, A.S. and Belyakova, L.E. (2002), Current Opinion in Colloid and Interface Science 7, 428-444.
- Tolstoguzov (2006), Biotechnology Advances 24:626-628.

Thank you very much for your attention.

Do you have any questions?

